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(54)	MESOGENIC COMPOUNDS,
	LIQUID-CRYSTALLINE MEDIA AND
	COMPONENTS FOR HIGH-FREQUENCY
	TECHNOLOGY

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(58) Field of Classification Search

See application file for complete search history.

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ABSTRACT (57)

A liquid-crystal medium which comprises a component A which consists of one or more compounds of the formula IA, in which the parameters have the respective meanings given in the claims or in the text, and to the corresponding, novel mesogenic compounds and to the preparation thereof. The use of these liquid-crystal media, in particular in components for high-frequency technology, and to components of this type which contain the media according to the invention, and to the production and use of these components. The components are suitable, in particular, for phase shifters in the microwave and millimeter wave region, for microwave and millimeter wave array antennae and very particularly for socalled tuneable "reflectarrays".

16 Claims, No Drawings

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MESOGENIC COMPOUNDS, LIQUID-CRYSTALLINE MEDIA AND COMPONENTS FOR HIGH-FREQUENCY TECHNOLOGY

FIELD OF THE INVENTION

The present invention relates to liquid-crystalline media, in particular for high-frequency technology, especially components for high-frequency devices, in particular antennas, especially for the gigahertz region and the terahertz region, which are operated in the microwave or millimeter wave region. These components use particular mesogenic, preferably liquid-crystalline, chemical compounds or liquid-crystalline media composed thereof for, for example, the phase

lateral methyl group on the central ring besides the strongly dielectrically positive isothiocyanatobistolan compounds of the formula

2

 C_4H_9 $C \equiv C$ $C \equiv C$ $C \equiv C$

Compounds containing three C—C triple bonds, such as, for example, the compound

$$C_5H_{11} - C = C - C_5H_{11},$$

shifting of microwaves for tuneable phased-array antennas or for tuneable cells of microwave antennas based on "reflectarrays". In addition, the present invention relates to novel mesogenic compounds. are mentioned in JP 2003-207631 A) and are proposed for use in optical films, polarisers and in liquid crystals of the light-scattering type.

Compounds of the General Formula

PRIOR ART AND PROBLEM TO BE SOLVED

Liquid-crystalline media have long been utilised in electrooptical displays (liquid crystal displays—LCDs) in order to display information.

Bistolan compounds, also known as triphenyldiacetylenes, $_{40}$ having an additional alkyl substitution on the central phenylene ring are adequately known to the person skilled in the art.

For example, Wu, S.-T., Hsu, C.-S, and Shyu, K.-F., Appl. 45 Phys. Lett., 74 (3), (1999), pages 344-346, discloses various liquid-crystalline bistolan compounds containing a lateral methyl group, of the formula

$$Ra + \underbrace{\begin{pmatrix} A^1 \end{pmatrix}}_{R} Z^1 + \underbrace{\begin{pmatrix} A^1 \end{pmatrix}}_{Im} R$$

in which

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$$C_nH_{2n+1}$$
 $C \equiv C$ CH_3 $C \equiv C$ C_mH_{2m+1}

Besides liquid-crystalline bistolan compounds of this type containing a lateral methyl group, Hsu, C. S., Shyu, K. F., Chuang, Y. Y. and Wu, S.-T., Liq. Cryst., 27 (2), (2000), pages 283-287, also discloses corresponding compounds containing a lateral ethyl group and proposes the use thereof, inter alia, in liquid crystal optically phased arrays.

Dabrowski, R., Kula, P., Gauza, S., Dziadiszek, J., Urban, 65 S, and Wu, S.-T., IDRC 08, (2008), pages 35-38, mentions dielectrically neutral bistolan compounds with and without a

denotes, for example,

are described in EP 1 655 360 A1.

However, liquid-crystalline media have recently also been proposed for use in components for microwave technology, as described, for example, in DE 10 2004 029 429 A and in JP 2005-120208 (A).

An industrially valuable application of liquid-crystalline media in high-frequency technology is based on their property that their dielectric properties can be controlled, particularly for the gigahertz region and the terahertz region, by a variable voltage. This enables the construction of tuneable antennas which contain no moving parts (Gaebler, A., Moessinger, A., Goelden, F., et al., "Liquid Crystal-Reconfigurable Antenna Concepts for Space Applications at Microwave and Millimeter Waves", International Journal of Antennas and Propagation, Volume 2009, Article ID 876989, 15 (2009), pages 1-7, DOI: 10.1155/2009/876989).

Penirschke, A., Müller, S., Scheele, P., Weil, C., Wittek, M., Hock, C. and Jakoby, R.: "Cavity Perturbation Method for Characterisation of Liquid Crystals up to 35 GHz", 34th European Microwave Conference—Amsterdam, pp. 545-548, describe, inter alia, the properties of the known single liquid-crystalline substance K15 (also called 4-n-pentyl-4'-cyanobi-phenyl or PP-5-N, Merck KGaA, Germany) at a frequency of 9 GHz.

DE 10 2004 029 429 A describes the use of liquid-crystal media in microwave technology, inter alia in phase shifters. DE 10 2004 029 429 A has already investigated liquid-crystalline media with respect to their properties in the corresponding frequency range.

For use in high-frequency technology, liquid-crystalline media having particular, hitherto rather unusual, unconventional properties, or combinations of properties, are required.

A. Gaebler, F. Goelden, S. Müller, A. Penirschke and R. 35 Jakoby "Direct Simulation of Material Permittivites using an Eigen-Susceptibility Formulation of the Vector Variational Approach", 12MTC 2009—International Instrumentation and Measurement Technology Conference, Singapore, 2009 (IEEE), pp. 463-467, describe the corresponding properties of the known liquid-crystal mixture E7 (likewise Merck KGaA, Germany).

DE 10 2004 029 429 A describes the use of liquid-crystal media in microwave technology, inter alia in phase shifters. 45 DE 10 2004 029 429 A has already investigated liquid-crystalline media with respect to their properties in the corresponding frequency range. In addition, it mentions liquid-crystalline media which comprise compounds of the formulae

$$C_nH_{2n+1}$$
 — CN

besides compounds of the formulae

$$C_nH_{2n+1}$$
 $C \equiv C$ and

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-continued
$$C_nH_{2n+1} \longrightarrow C \equiv C \longrightarrow NCS$$

Liquid-crystal media which comprise, for example, compounds of the formula

$$C_{n}H_{2n+1} \longrightarrow C \equiv C \longrightarrow C_{m}H_{2m+1}$$

are proposed for use in components for high-frequency technology, for example, in A. Lapanik, "Single compounds and mixtures for microwave applications, Dielectric, microwave studies on selected systems", dissertation, Technical University of Darmstadt, 2009, (D17).

However, the compositions known to date are afflicted with serious disadvantages. Besides other deficiencies, most of them result in disadvantageously high losses and/or inadequate phase shift or inadequate material quality (η) .

Novel liquid-crystalline media having improved properties are thus necessary. In particular, the loss in the microwave region and/or millimeter wave region must be reduced and the material quality improved.

In addition, there is a demand for an improvement in the low-temperature behaviour of the liquid-crystalline media and thus also of the components. Both an improvement in the operating properties and also in the shelf life are necessary here.

Thus, there is a considerable demand for liquid-crystalline media having suitable properties for corresponding practical applications.

PRESENT INVENTION

Surprisingly, it has now been found that it is possible to achieve components for high-frequency technology which do not have the disadvantages of the prior-art materials, or at least only do so to a considerably reduced extent, if selected liquid-crystalline media are employed.

The present invention thus relates to liquid-crystalline media which comprise one or more compounds of the formula IA,

$$R^{11}$$
 $C \equiv C$ A^{1} $C \equiv C$ R^{12}

60 in which

$$\left\langle A^{I}\right\rangle -$$

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very particularly preferably

denotes

$$\begin{array}{c}
 & \text{V}^{11} \\
 & \text{R}^{13} \\
 & \text{Or} \\
 & \text{V}^{12}
\end{array}$$
or
$$\begin{array}{c}
 & \text{R}^{13} \\
 & \text{I}^{1}
\end{array}$$

preferably

particularly preferably

$$\operatorname{CH_3}$$
 $\operatorname{CH_3}$
 $\operatorname{CH_3}$
 $\operatorname{CH_3}$
 $\operatorname{CH_3}$

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 $\begin{array}{c} L^1 \ denotes \ alkyl \ having \ 1 \ to \ 6 \ C \ atoms, \ cycloalkyl \ having \ 3 \ to \\ 6 \ C \ atoms, \ cycloalkenyl \ having \ 4 \ to \ 6 \ C \ atoms \ or \ halogen, \\ preferably \ CH_3, \ C_2H_5, \ n-C_3H_7(\ (-(CH_2)_2CH_3), i-C_3H_7(\ (-(CH_2)_2CH_3), i-C_3H_7(\ (-(CH_3)_2), \ cyclopropyl, \ cycloputyl, \ cyclopentyl, \ cyclopentyl, \ cyclopentyl, \ cyclopentyl, \ cyclopentyl, \ cyclopentyl, \ cyclopentyl \ preferably \ CH_3, \ C_2H_5, \ cyclopropyl, \ cyclobutyl \ or \ F \ and \ very \ particularly \ preferably \ F, \end{array}$

Y¹¹ and Y¹², independently of one another, denote H, alkyl having 1 to 6 C atoms, cycloalkyl having 3 to 6 C atoms, cycloalkenyl having 4 to 6 C atoms or halogen, preferably H, CH₃, C₂H₅, n-C₃H₇ (—(CH₂)₂CH₃), i-C₃H₇ (—CH (CH₃)₂), cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl, cyclopent-1-enyl, cyclohex-1-enyl, F or Cl and particularly preferably H, CH₃, C₂H₅ or F and very particularly preferably H or CH₃,

R¹¹ to R¹³, independently of one another, denote unfluorinated alkyl or unfluorinated alkoxy, each having 1 to 15 C atoms, unfluorinated alkenyl, unfluorinated alkenyloxy or unfluorinated alkoxyalkyl, each having 2 to 15 C atoms, or cycloalkyl, alkylcycloalkyl, cycloalkenyl, alkylcycloalkenyl, alkylcycloalkyl or alkylcycloalkenylalkyl, each having up to 15 C atoms, preferably

40 R¹¹ and R¹², independently of one another, denote unfluorinated alkyl or unfluorinated alkoxy, each having 1 to 7 C atoms, or unfluorinated alkenyl, unfluorinated alkenyloxy or unfluorinated alkoxyalkyl, each having 2 to 7 C atoms, particularly preferably

45 R¹¹ denotes unfluorinated alkyl having 1 to 7 C atoms or unfluorinated alkenyl, unfluorinated alkenyloxy or unfluorinated alkoxyalkyl, each having 2 to 7 C atoms, and particularly preferably

R¹² denotes unfluorinated alkyl or unfluorinated alkoxy, each having 1 to 7 C atoms, and preferably

R¹³ denotes unfluorinated alkyl having 1 to 5 C atoms, unfluorinated cycloalkyl or cycloalkenyl having 3 to 7 C atoms, unfluorinated alkylcyclohexyl or unfluorinated cyclohexylalkyl, each having 4 to 12 C atoms, or unfluorinated alkylcyclohexylalkyl having 5 to 15 C atoms, particularly preferably cyclopropyl, cyclobutyl, cyclopentyl or cyclohexyl, and very particularly preferably R¹³ denotes n-alkyl, particularly preferably methyl, ethyl or n-propyl, and the other denotes H or n-alkyl, particularly preferably H, methyl, ethyl or n-propyl.

The liquid-crystal media according to the invention are eminently suitable for use in components for high-frequency technology or for the microwave region and/or millimeter wave region of the electromagnetic spectrum. The present invention relates to this use of the media and to these compo-

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In a first preferred embodiment of the present invention, the component for high-frequency technology contains a liquid-crystal medium which comprises a component A which consists of one compound, two compounds or a plurality of compounds of the formula IA.

In accordance with a further preferred embodiment of the present invention, the component for high-frequency technology contains a liquid-crystalline medium comprising

a first component, component A, which comprises one or more compounds of the above-mentioned formula IA 10 and which optionally comprises one or more compounds of the formula D3 given below and

one or more further components selected from the group of components B to E defined below,

a strongly dielectrically positive component, component 15 B, which has a dielectric anisotropy of 10.0 or more,

a strongly dielectrically negative component, component C, which has a dielectric anisotropy of -5.0 or less,

a further component, component D, which has a dielectric anisotropy in the range from more than -5.0 to less than 20 10.0 and consists of compounds having seven or more, preferably eight or more, five- or six-membered rings, and

a further component, component E, which likewise has a dielectric anisotropy in the range from more than -5.0 to 25 less than 10.0 and consists of compounds having up to six five- or six-membered rings.

Typical examples of five-membered rings are

and others.

Typical examples of six-membered rings are

$$, \qquad \downarrow F, \qquad \downarrow F$$

The five- and six-membered rings also include saturated and partially saturated rings, as well as heterocyclic rings.

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For the purposes of the present application, condensed ring systems which consist of two of these rings, i.e. two five-membered rings, one five-membered ring or two six-membered rings, such as, for example,

are counted as one of these five- or six-membered rings on assignment of the compounds to components A or D.

Correspondingly, condensed ring systems which consist of a combination of three or more of these rings which are incorporated into the molecule in the longitudinal direction, such as, for example,

are counted as two of these five- or six-membered rings.

By contrast, condensed ring systems which are incorporated into the molecule in the transverse direction, such as, for example,

IA-4

are counted as one of these five- or six-membered rings.

The present invention likewise relates to the directly preceding liquid-crystalline media and to those described below, and to the use thereof in electro-optical displays and in particular in components for high-frequency technology.

In a preferred embodiment of the present invention, the liquid-crystal medium comprises one or more compounds of the formula IA, preferably selected from the group of the compounds of the formulae IA-1 to IA-4, preferably of the formulae IA-1 and/or IA-2:

-continued

in which the parameters have the respective meanings indicated above for formula IA, and preferably

R¹³ denotes alkyl having 1 to 6 C atoms, alkenyl having 2 to 6 C atoms, cycloalkyl having 3 to 6 C atoms or cycloalkenyl having 4 to 6 C atoms, preferably CH₃, C₂H₅, n-C₃H₇ (—(CH₂)₂CH₃), i-C₃H₇ (—CH(CH₃)₂), —CH—CH₂, cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl, cyclopent-1-enyl or cyclohex-1-enyl, and particularly preferably CH₃, C₂H₅, cyclopropyl or cyclobutyl,

L¹ denotes alkyl having 1 to 3 C atoms or halogen, preferably H, F or Cl, and particularly preferably H, F or CH₃, even more preferably H or F and very particularly preferably F,

²⁵ R¹¹ denotes unfluorinated alkyl having 1 to 7 C atoms, and

 R^{12} denotes unfluorinated alkyl having 1 to 7 C atoms or unfluorinated alkoxy having 1 to 7 C atoms.

In a preferred embodiment of the present invention, component A consists predominantly, even more preferably consists essentially and very particularly preferably consists completely of compounds of the formula IA.

In a particularly preferred embodiment of the present invention, the liquid-crystal medium comprises one or more compounds of the formula IA-1, preferably selected from the group of the compounds of the formulae IA-1a to IA-1e and IA-2a to IA-2e, particularly preferably selected from the group of the compounds of the formulae IA-1a to IA-1c, IA-1e, IA-2a to IA-2c and IA-2e, very particularly preferably selected from the group of the compounds of the formulae IA-1a, IA-1b, IA-2a and IA-2b,

$$R^{11}$$
 $C \equiv C$
 R^{13}
 $C \equiv C$
 R^{12}
 $C \equiv C$
 $C \equiv C$
 $C \equiv C$
 $C \equiv C$

$$R^{11} - C = C - C - R^{13}$$

$$R^{11}$$
 $C \equiv C$ R^{13} $C \equiv C$ R^{12} $C \equiv C$

$$R^{11} - C \equiv C - C = C$$

$$C = C$$

$$F$$

$$C = C$$

$$IA-1a$$

$$IA-1a$$

$$IA-1a$$

$$R^{11} - C \equiv C - CH_3$$

$$C \equiv C - CH_3$$

$$IA-1c$$

$$R^{11}$$
 $C \equiv C$ CH_3 $C \equiv C$ R^{12}

-continued

IA-1d ·CH₃ CH₃ 10 IA-1e ·CH₃ CH₃ IA-2a ₂₀ 25 IA-2b IA-2c 35 40 IA-2d CH₃ IA-2e ₅₀ CH₃ 55 Н3С

in which the parameters have the meanings given above under formula IA-1 or under formula IA-2, and preferably

 R^{11} and $R^{12},$ independently of one another, denote an alkyl radical having 2 to 7 C atoms, for example a propyl radical and a hexyl radical or in each case a propyl, butyl, pentyl or hexyl radical.

The liquid-crystal medium in component A optionally comprises one or more compounds of the formula IB,

$$\mathbb{R}^{13}$$
 \mathbb{R}^{14}
 \mathbb{R}^{14}

preferably

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13 -continued -CH₃ CH₃-CH₃-CH₃

-continued CH₃ particularly preferably -CH₃,

-continued

F

CH₃ R^{13} R^{14} R^{14} R^{14} R^{14} R^{14} R^{14} R^{14} R^{14}

 L^1 denotes alkyl having 1 to 6 C atoms, cycloalkyl having 3 to 6 C atoms or cycloalkenyl having 4 to 6 C atoms, preferably CH₃, C₂H₅, n-C₃H₇ (—(CH₂)₂CH₃), i-C₃H₇ (—CH (CH₃)₂), cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl,

preferably R¹¹

and R¹², independently of one another, denote unfluorinated alkyl or unfluorinated alkoxy, each having 1 to 7 C atoms, or unfluorinated alkenyl, unfluorinated alkenyloxy or unfluorinated alkoxyalkyl, each having 2 to 7 C atoms, particularly

preferably R¹¹ denotes unfluorinated alkyl having 1 to 7 C atoms or unfluorinated alkenyl, unfluorinated alkenyloxy or unfluorinated alkoxyalkyl, each having 2 to 7 C atoms, and particularly

preferably R¹² denotes unfluorinated alkyl or unfluorinated alkoxy, each having 1 to 7 C atoms, and preferably R¹³ and R¹⁴ denote H, unfluorinated alkyl having 1 to 5 C atoms, unfluorinated cycloalkyl or cycloalkenyl having 3 to 7 C atoms, unfluorinated alkylcyclohexyl or unfluorinated cyclohexylalkyl, each having 4 to 12 C atoms, or unfluorinated alkylcyclohexylalkyl having 5 to 15 C atoms, particularly preferably cyclopropyl, cyclobutyl, cyclopentyl or cyclohexyl, and very particularly preferably at least one of R¹³ and R¹⁴ denotes n-alkyl, particularly preferably methyl, ethyl or n-propyl, and the other denotes H or n-alkyl, particularly preferably H, methyl, ethyl or n-propyl.

In a particularly preferred embodiment of the present invention, component A comprises one or more compounds of the formula IA and one or more compounds of the formula IB, preferably component A consists predominantly, even more preferably consists essentially and very particularly preferably consists completely of compounds of the formulae IA and IB.

The compounds of the formula IB are preferably compounds selected from the group of the compounds of the formulae IB-1 to IB-4

$$R^{11} \longrightarrow C \equiv C \longrightarrow C \equiv C \longrightarrow R^{12}$$

$$R^{13} \longrightarrow C \equiv C \longrightarrow R^{14}$$

$$R^{14} \longrightarrow C \equiv C \longrightarrow R^{12}$$

$$R^{13} \longrightarrow C \equiv C \longrightarrow R^{12}$$

$$R^{14} \longrightarrow C \equiv C \longrightarrow R^{12}$$

cyclopent-1-enyl or cyclohex-1-enyl, and particularly preferably ${\rm CH_3}, {\rm C_2H_5},$ cyclopropyl or cyclobutyl,

X¹ denotes H, alkyl having 1 to 3 C atoms or halogen, preferably H, F or Cl, particularly preferably H or F and very particularly preferably F,

R¹¹ to R¹⁴, independently of one another, denote unfluorinated alkyl or unfluorinated alkoxy, each having 1 to 15 C 60 atoms, unfluorinated alkenyl, unfluorinated alkenyloxy or unfluorinated alkoxyalkyl, each having 2 to 15 C atoms, or cycloalkyl, alkylcycloalkyl, cycloalkenyl, alkylcycloalkenyl, alkylcycloalkylalkyl or alkylcycloalkenylalkyl, each having up to 15 C atoms, and alternatively one of R¹³ and R¹⁴ or both also denote H,

in which

L¹ denotes alkyl having 1 to 6 C atoms, alkenyl having 2 to 6 C atoms, cycloalkyl having 3 to 6 C atoms or cycloalkenyl having 4 to 6 C atoms, preferably $\mathrm{CH_3}$, $\mathrm{C_2H_5}$, $\mathrm{n\text{-}C_3H_7}(-(\mathrm{CH_2})_2\mathrm{CH_3})$, $\mathrm{i\text{-}C_3H_7}(-\mathrm{CH}(\mathrm{CH_3})_2)$, $-\mathrm{CH}=-\mathrm{CH_2}$, cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl, cyclopent-1-enyl or cyclohex-1-enyl, and particularly preferably $\mathrm{CH_3}$, $\mathrm{C_2H_5}$, cyclopropyl or cyclobutyl,

X¹ denotes H, alkyl having 1 to 3 C atoms or halogen, preferably H, F or Cl, particularly preferably H, F or CH₃, even more preferably H or F and very particularly preferably F, and the other parameters have the respective meanings indicated above for formula IB, and preferably

R¹¹ denotes unfluorinated alkyl having 1 to 7 C atoms, and

 R^{12} denotes unfluorinated alkyl having 1 to 7 C atoms or unfluorinated alkoxy having 1 to 7 C atoms.

In a particularly preferred embodiment of the present invention, the liquid-crystal medium comprises one or more compounds of the formula IB-1, preferably selected from the group of the compounds of the formulae IB-1a-1 to IB-1a-12 and IB-1b-1 to IB-1b-12

10 IB-1a-1

$$R^{11}$$
 $C \equiv C$ $C \equiv C$ R^{12} $C \equiv C$

IB-1a-2

IB-1a-3 25

IB-1a-4

IB-1a-5

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$$R^{11}$$
 $C \equiv C$ $C \equiv C$ R^{12}

 \mathbb{R}^{11} \mathbb{C} \mathbb{C} \mathbb{C} \mathbb{C} \mathbb{R}^{12}

$$\mathbb{R}^{11}$$
 \mathbb{C} \mathbb{C}

$$R^{11}$$
 $C \equiv C$ $C \equiv C$ R^{12} $C \equiv C$ $C \equiv C$ $C \equiv C$

$$R^{11}$$
 $C \equiv C$ $C \equiv C$ R^{12} 50 IB-1a-7

$$R^{11}$$
 $C \equiv C$ $C \equiv C$ R^{12} $C \equiv C$ $C \equiv C$

$$\mathbb{R}^{11}$$
 \mathbb{C} $\mathbb{E}\mathbb{C}$ \mathbb{C} $\mathbb{E}\mathbb{C}$ \mathbb{R}^{12} 65

-continued

$$R^{11}$$
 $C \equiv C$ $C \equiv C$ R^{12} $C \equiv C$

$$R^{11}$$
 $C \equiv C$ $C \equiv C$ R^{12} $C \equiv C$

$$R^{11} \longrightarrow C \equiv C \longrightarrow C \equiv C \longrightarrow R^{12}$$
IB-1a-12

$$R^{11} - C \equiv C - C \equiv C - R^{12}$$

$$IB-1b-1$$

$$R^{11}$$
 $C \equiv C$ $C \equiv C$ R^{12} $B-1b-2$

$$R^{11}$$
 $C \equiv C$ $C \equiv C$ R^{12} $IB-1b-3$

$$R^{11}$$
 $C \equiv C$ $C \equiv C$ R^{12} $C \equiv C$

$$R^{11}$$
 $C \equiv C$ $C \equiv C$ R^{12} $B-1b-5$

$$R^{11}$$
 $C \equiv C$ $C \equiv C$ R^{12}

-continued IB-1b-6 IB-1b-7 10 IB-1b-8 ₁₅ IB-1a-9 IB-1b-10 ₃₀

$$R^{11} - C \equiv C - C \equiv C - R^{12}$$

-continued

IB-1b-12

$$R^{11}$$
 $C \equiv C$ $C \equiv C$ R^{12}

in which the parameters have the meanings given above under

formula IB-1, and preferably $$\rm R^{11}$ and $\rm R^{12}$, independently of one another, denote an alkyl radical having 2 to 7 C atoms, for example a propyl radical and a hexyl radical or in each case a propyl, butyl, pentyl or hexyl radical.

In a very particularly preferred embodiment of the present invention, the liquid-crystal medium, or component A of the liquid-crystal medium, comprises one or more compounds of the formula IB, preferably selected from the group of the compounds of the formulae IB-1a-2, IB-1a-5, IB-1a-7, $IB\text{-}1a\text{-}8,\ IB\text{-}1a\text{-}9,\ IB\text{-}1a\text{-}10,\ IB\text{-}1b\text{-}5,\ IB\text{-}1b\text{-}7,\ IB\text{-}1b\text{-}8,$ IB-1b-9, IB-1b-10, where the parameters have the meanings given above, and particularly preferably

R¹¹ and R¹², independently of one another, denote unfluorinated alkyl having 1 to 7 C atoms or unfluorinated alkoxy having 1 to 6 C atoms,

especially preferably one of

R¹¹ and R¹² denotes alkyl and the other denotes alkyl or alkoxy, and very particularly preferably R¹¹ and R¹² have different meanings from one another.

In a preferred embodiment of the present invention, the liquid-crystal medium, or component A of the liquid-crystal medium, comprises one or more compounds of the formula IB-2, in which preferably

R¹¹ and R¹², independently of one another, denote an alkyl radical having 2 to 7 C atoms, for example a propyl radical and a hexyl radical or in each case a propyl, butyl, pentyl or hexyl radical.

In a preferred embodiment of the present invention, the liquid-crystal medium, or component A of the liquid-crystal medium, comprises one or more compounds of the formula IB-3, preferably selected from the group of the compounds of the formulae IB-3a-1 to IB-3a-3 and IB-3b-1 to IB-3b-3, preferably IB-3a-2, IB-3b-2,

$$R^{11} \longrightarrow C \equiv C \longrightarrow R^{12}$$

$$R^{11} \longrightarrow C \equiv C \longrightarrow R^{12}$$

$$R^{12} \longrightarrow R^{12}$$

$$R^{13} \longrightarrow C \equiv C \longrightarrow R^{12}$$

$$R^{14} \longrightarrow C \equiv C \longrightarrow R^{12}$$

$$R^{14} \longrightarrow C \equiv C \longrightarrow R^{12}$$

-continued

IB-3b-1

$$R^{11}$$
 $C \equiv C$
 R^{12}

IB-3b-2

 R^{11}
 $C \equiv C$
 R^{12}

IB-3b-3

in which the parameters have the meanings given above under

formula IB-3, and preferably

R¹¹ and R¹², independently of one another, denote an alkyl radical having 2 to 7 C atoms, for example a propyl radical and a hexyl radical or in each case a propyl, butyl, pentyl or hexyl radical.

In a preferred embodiment of the present invention, the liquid-crystal medium, or component A of the liquid-crystal medium, comprises one or more compounds of the formula IB-4, preferably selected from the group of the compounds of the formulae IB-4-a-1 to IB-4-a-3 and IB-4-b-1 to IB-4-b-3, preferably IB-4-b-2,

$$R^{11} \longrightarrow C \equiv C \longrightarrow C \equiv C \longrightarrow R^{12}$$

$$R^{11} \longrightarrow C \equiv C \longrightarrow C \equiv C \longrightarrow R^{12}$$

$$R^{11} \longrightarrow C \equiv C \longrightarrow C \equiv C \longrightarrow R^{12}$$

$$R^{11} \longrightarrow C \equiv C \longrightarrow C \equiv C \longrightarrow R^{12}$$

$$R^{11} \longrightarrow C \equiv C \longrightarrow C \equiv C \longrightarrow R^{12}$$

$$R^{11} \longrightarrow C \equiv C \longrightarrow C \equiv C \longrightarrow R^{12}$$

$$R^{11} \longrightarrow C \equiv C \longrightarrow C \equiv C \longrightarrow R^{12}$$

$$R^{11} \longrightarrow C \equiv C \longrightarrow C \equiv C \longrightarrow R^{12}$$

$$R^{11} \longrightarrow C \equiv C \longrightarrow C \equiv C \longrightarrow R^{12}$$

$$R^{12} \longrightarrow C \equiv C \longrightarrow C \equiv C \longrightarrow R^{12}$$

in which the parameters have the meanings given above under formula IB-4, and preferably

 R^{11} and $R^{12},\,$ independently of one another, denote an alkyl radical having 2 to 7 C atoms, for example a propyl radical

and a hexyl radical or in each case a propyl, butyl, pentyl or hexyl radical.

The compounds of the formula IA can advantageously be prepared as evident from the following illustrative synthesis (Schemes 1 and 2):

Scheme 1: Illustrative synthesis of compounds of the formula IA-1 (for example IA-1b).

Scheme 2: Illustrative synthesis of the compounds of the formula IA-2 (fore example IA-2a).

$$\begin{array}{c|c} F & & \\ \hline \\ & &$$

$$\mathbb{R}^{12} \longrightarrow \mathbb{R}^{11} \xrightarrow{\begin{array}{c} Cu(I) \\ Pd(0) \end{array}} \mathbb{R}^{11}$$

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The compounds of the formula IB-1a can advantageously be prepared as evident from the following illustrative synthesis (Schemes 3-6):

Scheme 3: Illustrative synthesis of the compounds of the formula IB-1 (for example IB-1a-7 to IB-1a-12) containing symmetrical end groups, where cycloalkyl also denotes cycloalkenyl.

$$Cl \xrightarrow{C_{3.6} \text{cycloalkyl-B(OH)}_2} Cl \xrightarrow{C_{3.6} \text{cycloalkyl-B(OH)}_2} Cl \xrightarrow{\text{cycloalkyl-B(OH)}_2} R^{11} \text{ Pd cat.}$$

Scheme 4: Illustrative synthesis of the compounds of the formula IB-1 (for example IB-1a-7, IB-1a-8, IB-1a-9 and IB-1a-11) containing symmetrical end groups.

Scheme 5: Illustrative synthesis of the compounds of the formula IB-1 (for example IB-1a-7 to IB-1a-12) containing asymmetrical end groups, where cycloalkyl also denotes cycloalkenyl.

$$C_1$$
 OH C_{3-6} cycloalkyl-B(OH)₂ Pd cat.

-continued cycloalkyl

CI

OH $(Tf)_2O$ OH R^{12} Pd cat. R^{12} R^{12} R^{12} R^{12} R^{12} R^{12} R^{12} R^{12} R^{12}

30 Scheme 6: Illustrative synthesis of the compounds of the formula IB-1 (for example IB-1a-7 to IB-1a-12) containing asymmetrical end groups, where cycloalkyl also denotes cycloalkenyl.

$$\begin{array}{c|c} C_{3\text{-6cycloalkyl-B(OH)_2}} \\ \hline \\ Cl \end{array} \begin{array}{c} C_{3\text{-6cycl$$

The compounds of the formula IB-1b can advantageously be obtained in accordance with the following general reaction schemes (Reaction Schemes 5 to 11). The parameters L^1, R^{11}, R^{12} and X^1 therein are as defined above and below. R has the meaning of R^{11} or $R^{12}.$

Scheme 7: Illustrative synthesis of the compounds of the formula IB with asymmetrical substitution, where R has the meaning of R^{11} .

$$R \longrightarrow H$$

$$OMe \begin{vmatrix} 1. & \text{Li} \longrightarrow \text{N}(\text{Si}(\text{CH}_3)_3)_2 \\ 2. & \text{CH}_3\text{O} \longrightarrow \text{BBN} \end{vmatrix}$$

$$3. \qquad L^1 \qquad X^1$$

$$1/2 \qquad Br \longrightarrow Br \qquad 25$$

$$4. & ((\text{CyHex})_3P)_2PdCl_2, \text{ base}$$

$$R \longrightarrow R$$

Scheme 8: Illustrative synthesis of the compounds of the formula IB with asymmetrical substitution.

Schemes 9 to 13 show the synthesis of variously substituted central rings. The phenylalkynyl radicals here can be generalised to any desired substituted phenylalkynyl radicals.

Scheme 9: Illustrative synthesis of the compounds of the formula IB-1 (for example IB-1b-1 to IB-1b-4) containing symmetrical end groups, where R has the meaning of \mathbb{R}^{11} .

Scheme 10: Illustrative synthesis of the compounds of the formula IB-1b-8 containing symmetrical end groups, where R has the meaning of R^{11} .

Scheme 11: Illustrative synthesis of the compounds of the formula IB-1 where X^1 = Cl, L^1 = CH_3 and containing symmetrical end groups, where R has the meaning of R^{11} .

CI

CH₃

OH

$$\downarrow Bu_4NBr_3$$

CI

CH₃

OH

$$\downarrow (CF_3SO_2)_2O, base$$

CI

CH₃

OSO₂CF₃

$$\downarrow ((CyHex)_3P)_2PdCl_2, base$$

CI

CH₃

CH₃
 $\downarrow ((CyHex)_3P)_2PdCl_2, base$

CI

CH₃

R

CI

CH₃

R

Scheme 12: Illustrative synthesis of the compounds of the formula IB where $X^1 = L^1 = CH_3$ and containing symmetrical end groups, where R has the meaning of R^{11} .

$$\begin{array}{c} \text{H}_3\text{C} \\ \text{CH}_3 \\ \text{OH} \\ \\ \text{(CF}_3\text{SO}_2)_2\text{O, base} \end{array}$$

$$Br \longrightarrow OSO_2CF_3$$

$$0SO_2CF_3$$

$$1. 2 \text{ Li} \longrightarrow N(Si(CH_3)_3)_2$$

$$2. 2 \text{ CH}_3O \longrightarrow BBN$$

$$2 \text{ H} \longrightarrow R$$

$$((CyHex)_3P)_2PdCl_2, \text{ base}$$

Scheme 13: Illustrative synthesis of the compounds of the formula IB where $X^1 = L^1 =$ alkyl and containing symmetrical end groups, where R has the meaning of R^{11} and where the two alkyl groups "alkyl" may be identical to or different from one another and in addition one or both may be an alkenyl group.

$$\begin{array}{c|c} & Br & Cl & \\ & Li & TMP \\ & alkyl \cdot I & \\ & Br & alkyl \\ & Cl & Cl & \\ & & Alkyl \cdot B(OH)_2 \text{ or alkyl} \cdot BF_3K \\ & Pd \text{ cat base} \end{array}$$

-continued

The compounds of the formula IB-2 can advantageously be 25 prepared as evident from the following illustrative synthesis (Scheme 14):

Scheme 14: Illustrative synthesis of the compounds of the formula IB-2 containing asymmetrical end groups, where R is defined corresponding to R^{11} and R^{\prime} is defined corresponding to R^{12} .

Br
$$\frac{\text{BuLi, I}_{2, \text{THF}}}{-70^{\circ} \text{C.}}$$

Br $\frac{\text{BuLi, I}_{2, \text{THF}}}{\text{Cu(I)/Pd(0)}}$

NEt₃

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-continued

$$R'$$
 $Cu(I)/Pd(0)$
 NEt_3
 R'
 R'
 R

The compounds of the formulae IB-3 and IB-4 can advantageously be obtained in accordance with the following general reaction scheme (Reaction Scheme 15).

Scheme 15: Illustrative synthesis of the compounds of the formulae IB-3 and IB-4 containing symmetrical end groups, in which R has the meaning of \mathbb{R}^{11} or \mathbb{R}^{12} .

-continued

Br
$$R^{13}$$
 R^{13}
 R^{13}

Besides component A, these media according to the invention preferably comprise a component selected from the two components B and C and optionally additionally component D and/or component E.

These media according to the invention preferably comprise two, three or four, particularly preferably two or three, components selected from the group of components A to E. These media preferably comprise

component A and component B, or

component A, component B and component D and/or E, or component A and component C, or

component A, component B and component C, or component A, component C and component D and/or E.

These media according to the invention preferably comprise a component B and no component C or vice versa.

The strongly dielectrically positive component, component B, preferably has a dielectric anisotropy of 20.0 or more, more preferably 25.0 or more, particularly preferably 30.0 or more and very particularly preferably 40.0 or more.

The strongly dielectrically negative component, component C, preferably has a dielectric anisotropy of -7.0 or less, more preferably -8.0 or less, particularly preferably -10.0 or less and very particularly preferably -15.0 or less.

In a preferred embodiment of the present invention, component B comprises one or more compounds selected from the group of the compounds of the formulae IIA and IIB:

 $\mathbb{R}^{21} \xrightarrow{F} \mathbb{CF}_2 \text{-O} \xrightarrow{F} \mathbb{X}^2$ \mathbb{IIA} \mathbb{F} \mathbb{F}

$$\mathbb{R}^{21}$$
 $+$ \mathbb{R}^{21} $+$ \mathbb{R}^{21} $+$ \mathbb{R}^{22} $+$ \mathbb{R}^{22}

R²¹ denotes unfluorinated alkyl or unfluorinated alkoxy, each having 1 to 15 C atoms, or unfluorinated alkenyl, unfluorinated alkenyloxy or unfluorinated alkoxyalkyl, each having 2 to 15 C atoms, preferably alkyl, particularly preferably n-alkyl,

R²² denotes H, unfluorinated alkyl or unfluorinated alkoxy, each having 1 to 5, preferably 1 to 3, particularly preferably 3, C atoms,

$$A^{21}$$
 to A^{22} ,

independently of one another and, if they occur more than once, these also in each case independently of one another, denote

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preferably

n and m, independently of one another, denote 1 or 2, preferably

(n+m) denotes 3 or 4, and particularly preferably n denotes 2,

X² denotes F, Cl, —CF₃ or —OCF₃, preferably F or Cl,

particularly preferably F, Y² denotes F, Cl, —CF₃, —OCF₃ or CN, preferably CN, and ₂₅ Z^2 denotes H or F.

Preferred compounds of the formula IIA are the compounds of the corresponding sub-formula IIA-1

$$\mathbb{R}^{21} - \mathbb{Q} + \mathbb{$$

in which R²¹ has the meaning given above.

Preferred compounds of the formula IIB are the compounds of the corresponding sub-formulae IIB-1 and IIB-2:

in which R²¹, R²² and X² have the respective meanings given

In a preferred embodiment of the present invention, component C comprises one or more compounds selected from the group of the compounds of the formulae IIIA and IIIB:

$$R^{31}$$
 $C \equiv C$ F F F F R^{32}

IIIA

in which

 R^{31} and R^{32} , independently of one another, have the meanings indicated above for R²¹ under formula IIA,

30 and preferably

$${f R}^{31}$$
 denotes ${f C}_n{f H}_{2n+1}$ or ${f CH}_2$ — ${f CH}$ — $({f CH}_2)_Z$ and ${f R}^{32}$ denotes ${f C}_m{f H}_{2m+1}$ or ${f O}$ — ${f C}_m{f H}_{2m+1}$ or $({f CH}_2)_Z$ — ${f CH}$ — ${f CH}_2$,

35 and in which

n and m, independently of one another, denote an integer in the range from 0 to 15, preferably in the range from 1 to 7 and particularly preferably 1 to 5, and

40 z denotes 0, 1, 2, 3 or 4, preferably 0 or 2.

The preferred combinations of (R31 and R32) here are, in particular, $(C_nH_{2n+1}$ and $C_mH_{2m+1})$ and (C_nH_{2n+1}) and $\mathrm{O}\text{---}\mathrm{C}_m\mathrm{H}_{2m+1}).$

Preferred compounds of the formula IIIB are the compounds of the sub-formulae IIIB-1 and IIIB-2:

$$\mathbb{R}^{21} - \mathbb{C}F_2 - \mathbb{O} - \mathbb{C}F_2 -$$

-continued

VI

VII

VIII

IX

$$C_nH_{2n+1} - C_mH_{2m+1}$$
IIIB-1

$$C_nH_{2n+1}$$
 O C_mH_{2m+1} O

in which

n and m each have the meanings given above for formula MB $\,^{20}$ and preferably, independently of one another, denote an integer in the range from 1 to 7.

In a preferred embodiment of the present invention, component D comprises one or more compounds of the following formula IV:

$$R^{41} = \begin{bmatrix} L^{41} & L^{42} \\ & & \\ L^{43} & L^{44} \end{bmatrix}$$

in which

 R^{41} and R^{42} , independently of one another, have one of the meanings indicated above for R^{11} under formula I,

 $\rm L^{41}$ to $\rm L^{44}$ on each appearance, in each case independently of 40 one another, denote H, alkyl having 1 to 5 C atoms, F or Cl, and

p denotes an integer in the range from 7 to 14, preferably from 8 to 12 and particularly preferably from 9 to 10, and preferably

at least two of the substituents

L⁴¹ to L⁴⁴ present have a meaning other than H, and

 R^{41} denotes C_nH_{2n+1} or CH_2 —CH— $(CH_2)_Z$, and

 R^{42} denotes C_mH_{2m+1} or $O-C_mH_{2m+1}$ or $(CH_2)_Z$ — $CH=CH_2$,

and in which

n and m, independently of one another, denote an integer in the range from 0 to 15, preferably in the range from 1 to 7 and particularly preferably 1 to 5, and

z denotes 0, 1, 2, 3 or 4, preferably 0 or 2.

In a preferred embodiment of the present application, the liquid-crystal medium additionally comprises a further component, component E, which preferably consists of one or more compounds selected from the group of the compounds of the formulae V to IX:

$$L^{51} \overbrace{\hspace{1cm}}^{51} A^{51} \overbrace{\hspace{1cm}}^{52} A^{52} \underbrace{\hspace{1cm}}^{52} L^{52} \qquad \qquad 65$$

 L^{71} A^{71} Z^{71} A^{72} Z^{72} A^{73} Z^{73} A^{74} L^{72}

$$R^{81}$$
 Z^{81} Z^{82} Z^{82} Z^{82} Z^{82} Z^{82}

$$L^{91}$$
 A^{91} Z^{91} A^{92} Z^{92} A^{93} Z^{93} A^{94} L^{92}

in which

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 L^{51} denotes R^{51} or X^{51} ,

 L^{52} denotes R^{52} or X^{52} ,

R⁵¹ and R⁵², independently of one another, denote H, unfluorinated alkyl or unfluorinated alkoxy having 1 to 17, preferably 3 to 10, C atoms or unfluorinated alkenyl, unfluorinated alkenyloxy or unfluorinated alkoxyalkyl having 2 to 15, preferably 3 to 10, C atoms, preferably alkyl or unfluorinated alkenyl,

X⁵¹ and X⁵², independently of one another, denote H, F, Cl, —CN, —NCS, —SF₅, fluorinated alkyl or fluorinated alkoxy having 1 to 7 C atoms or fluorinated alkenyl, unfluorinated or fluorinated alkenyloxy or unfluorinated or fluorinated alkoxyalkyl having 2 to 7 C atoms, preferably fluorinated alkoxy, fluorinated alkenyloxy, F or Cl, and

$$A^{51}$$
 to A^{53}

independently of one another, denote

preferably

 L^{61} denotes R^{61} and, in the case where Z^{61} and/or Z^{62} denote trans-CH=CH- or trans-CF=CF-, alternatively also 25

 L^{62} denotes R^{62} and, in the case where Z^{61} and/or Z^{62} denote trans-CH—CH— or trans-CF—CF—, alternatively also denotes X62.

 R^{61} and R^{62} , independently of one another, denote H, unflu- 30 orinated alkyl or unfluorinated alkoxy having 1 to 17, preferably 3 to 10, C atoms or unfluorinated alkenyl, unfluorinated alkenyloxy or unfluorinated alkoxyalkyl having 2 to 15, preferably 3 to 10, C atoms, preferably alkyl or unfluorinated alkenyl,

X⁶¹ and X⁶², independently of one another, denote F or Cl, —CN, —NCS, —SF $_5$, fluorinated alkyl or alkoxy having 1 to 7 C atoms or fluorinated alkenyl, alkenyloxy or alkoxyalkyl having 2 to 7 C atoms, preferably -NCS,

Z⁶¹ and Z⁶² denotes trans-CH—CH—, trans-CF—CF— or —C—C— and the other, independently thereof, denotes trans-CH=CH-, trans-CF=CF- or a single bond, preferably one of them denotes —C—C— or trans-CH—CH— and the other denotes a single bond, and

$$A^{61}$$
 to A^{63} ,

independently of one another, denote

-continued

$$F$$
 F
 F
 F
 F
 F

preferably

$$\underbrace{\hspace{1cm}}^F, \quad \underbrace{\hspace{1cm}}^F, \quad \underbrace{\hspace{1cm}}^F, \quad \underbrace{\hspace{1cm}}^F$$

 L^{71} denotes R^{71} or X^{71} ,

 L^{72} denotes R^{72} or X^{72} ,

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R⁷¹ and R⁷², independently of one another, denote H, unfluorinated alkyl or unfluorinated alkoxy having 1 to 17, preferably 3 to 10, C atoms or unfluorinated alkenyl, unfluorinated alkenyloxy or unfluorinated alkoxyalkyl having 2 to 15, preferably 3 to 10, C atoms, preferably alkyl or unfluorinated alkenyl,

 \boldsymbol{X}^{71} and $\boldsymbol{X}^{72},$ independently of one another, denote H, F, Cl, —CN, —NCS, —SF₅, fluorinated alkyl or fluorinated alkoxy having 1 to 7 C atoms or fluorinated alkenyl, unfluorinated or fluorinated alkenyloxy or unfluorinated or fluorinated alkoxyalkyl having 2 to 7 C atoms, preferably fluorinated alkoxy, fluorinated alkenyloxy, F or Cl, and

 Z^{71} to Z^{73} , independently of one another, denote trans-CH=CH-, trans-CF=CF-, -C=C- or a single bond, preferably one or more of them denote a single bond, particularly preferably all denote a single bond and

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$$A^{71}$$
 to A^{74} ,

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independently of one another, denote

preferably

R⁸¹ and R⁸², independently of one another, denote H, unfluorinated alkyl or alkoxy having 1 to 15, preferably 3 to 10, C atoms or unfluorinated alkenyl, alkenyloxy or alkoxyalkyl having 2 to 15, preferably 3 to 10, C atoms, preferably 45 unfluorinated alkyl or alkenyl,

one of

Z⁸¹ and Z⁸² denotes trans-CH=CH—, trans-CF=CF— or —C=C— and the other, independently thereof, denotes 50 trans-CH=CH—, trans-CF=CF— or a single bond, preferably one of them denotes —C=C— or trans-CH=CH— and the other denotes a single bond, and

$$A^{81}$$

denotes

-continued A^{83} ,

independently of one another, denote

 L^{91} denotes R^{91} or X^{91} ,

 L^{92} denotes R^{92} or X^{92} ,

³⁵ R⁹¹ and R⁹², independently of one another, denote H, unfluorinated alkyl or alkoxy having 1 to 15, preferably 3 to 10, C atoms or unfluorinated alkenyl, alkenyloxy or alkoxyalkyl having 2 to 15, preferably 3 to 10, C atoms, preferably unfluorinated alkyl or alkenyl,

40 X⁹¹ and X⁹², independently of one another, denote H, F, Cl,
—CN, —NCS, —SF₅, fluorinated alkyl or fluorinated alkoxy having 1 to 7 C atoms or fluorinated alkenyl, unfluorinated or fluorinated alkenyloxy or unfluorinated or fluorinated alkoxyalkyl having 2 to 7 C atoms, preferably fluorinated alkoxy, fluorinated alkenyloxy, F or Cl, and

Z⁹¹ to Z⁹³, independently of one another, denote trans-CH—CH—, trans-CF—CF—, —C—C— or a single bond, preferably one or more of them denotes a single bond, and particularly preferably all denote a single bond,

$$\left\langle A^{91}\right\rangle -$$

denotes

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V-3

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independently of one another, denote

and where compounds of the formula IIIA are excluded from the compounds of the formula ${\rm VI}$.

In a preferred embodiment of the present invention, the 35 liquid-crystal medium comprises, more preferably predominantly consists of, even more preferably essentially consists of and very particularly preferably completely consists of one or more compounds of the formula V, preferably selected from the group of the compounds of the formulae V-1 to V-3, preferably of the formulae V-1 and/or V-2 and/or V-3, preferably of the formulae V-1 and V-2:

$$R^{51}$$
 A^{51} A^{52} A^{53} X^{52} A^{53} A^{52} A^{53} A^{52}

in which the parameters have the respective meanings indicated above for formula V and preferably

R⁵¹ denotes unfluorinated alkyl having 1 to 7 C atoms or 60 unfluorinated alkenyl having 2 to 7 C atoms,

R⁵² denotes unfluorinated alkyl having 1 to 7 C atoms or unfluorinated alkenyl having 2 to 7 C atoms or unfluorinated alkoxy having 1 to 7 C atoms,

X⁵¹ and X⁵², independently of one another, denote F, Cl, 65 —OCF₃, —CF₃, —CN, —NCS or —SF_S, preferably F, Cl, —OCF₃ or —CN.

The compounds of the formula V-1 are preferably selected from the group of the compounds of the formulae V-1a to V-1d, more preferably these compounds of the formula V predominantly consist, even more preferably essentially consist and very particularly preferably completely consist thereof:

$$R^{51}$$
 F
 F
 Y^{51}
 F
 Y^{51}
 F
 Y^{51}
 Y^{52}
 Y^{51}
 Y^{52}
 Y^{52}
 Y^{51}
 Y^{52}
 Y^{52}

in which the parameters have the respective meanings indicated above for formula V-1 and in which

 Y^{51} and Y^{52} , in each case independently of one another, denote H or F, and preferably

R⁵¹ denotes alkyl or alkenyl, and

V-1 $_{45}$ X^{51} denotes F, Cl or —OCF₃.

The compounds of the formula V-2 are preferably selected from the group of the compounds of the formulae V-2a to V-2g and/or from the group of the compounds of the formulae V-2f and V-2g, more preferably these compounds of the formula V predominantly consist, even more preferably essentially consist and very particularly preferably completely consist thereof:

$$R^{51}$$
 F
 Y^{51}
 Y^{52}
 Y^{52}

where in each case the compounds of the formula V-2a are excluded from the compounds of the formulae V-2b and V-2c, the compounds of the formula V-2b are excluded from the compounds of the formula V-2c and the compounds of the 35 formula V-2e are excluded from the compounds of the formula V-2f, and

in which the parameters have the respective meanings indicated above for formula V-1 and in which

 Y^{51} and Y^{52} , in each case independently of one another, 40 denote H or F, and preferably

R⁵¹ denotes alkyl or alkenyl,

 X^{51} denotes F, Cl or $-OCF_3$, and preferably one of

 Y^{51} and Y^{52} denotes H and the other denotes H or F, preferably likewise denotes H.

The compounds of the formula V-3 are preferably compounds of the formula V-3a:

$$X^{51}$$
 X^{52}

in which the parameters have the respective meanings indicated above for formula V-1 and in which preferably X⁵¹ denotes F, Cl, preferably F,

X⁵² denotes F, Cl or —OCF₃, preferably —OCF₃.

In an even more preferred embodiment of the present invention, the compounds of the formula V are selected from the group of the compounds V-1a to V-1d, preferably selected from the group of the compounds V-1c and V-1d, more preferably these compounds of the formula V predominantly consist, even more preferably essentially consist and very particularly preferably completely consist thereof:

The compounds of the formula V-1a are preferably selected from the group of the compounds of the formulae V-1a-1 and V-1a-2, more preferably these compounds of the formula V predominantly consist, even more preferably essentially consist and very particularly preferably completely consist

$$\mathbb{R}^{51} \xrightarrow{F} \mathbb{F}$$

$$\mathbb{F}$$

$$\mathbb{F}$$

$$\mathbb{F}$$

$$\mathbb{F}$$

$$\mathbb{F}$$

$$\mathbb{F}$$

$$\mathbb{F}$$

$$\mathbb{R}^{51}$$
 \longrightarrow \mathbb{R}^{51} \longrightarrow \mathbb{R}^{51}

in which

R⁵¹ has the meaning indicated above and preferably denotes C_nH_{2n+1} , in which

n denotes an integer in the range from 0 to 7, preferably in the range from 1 to 5 and particularly preferably 3 or 7.

The compounds of the formula V-1b are preferably compounds of the formula V-1b-1:

$$\mathbb{R}^{51} \underbrace{\hspace{1.5cm}}^{F} F$$

in which

R⁵¹ has the meaning indicated above and preferably denotes C_nH_{2n+1} , in which

 V_{-3a} on denotes an integer in the range from 0 to 15, preferably in the range from 1 to 7 and particularly preferably 1 to 5.

> The compounds of the formula V-1c are preferably selected from the group of the compounds of the formulae V-1c-1 to V-1c-4, preferably selected from the group of the compounds of the formulae V-1c-1 and V-1c-2, more preferably these compounds of the formula V predominantly consist, even more preferably essentially consist and very particularly preferably completely consist thereof:

$$\mathbb{R}^{51}$$
 \mathbb{F} \mathbb{F}

$$\mathbb{R}^{51} \longrightarrow \mathbb{F}$$

$$\mathbb{R}^{51} \longrightarrow$$

in which

 R^{51} has the meaning indicated above and preferably denotes C_nH_{2n+1} , in which

n denotes an integer in the range from 0 to 15, preferably in the range from 1 to 7 and particularly preferably 1 to 5.

The compounds of the formula V-1d are preferably selected from the group of the compounds of the formulae V-1d-1 and V-1d-2, preferably the compound of the formula V-1d-2, more preferably these compounds of the formula V predominantly consist, even more preferably essentially consist and very particularly preferably completely consist thereof:

in which

 ${\bf R}^{51}$ has the meaning indicated above and preferably denotes ${\bf C}_n{\bf H}_{2n+1},$ in which

n denotes an integer in the range from 0 to 15, preferably in the range from 1 to 7 and particularly preferably 1 to 5.

The compounds of the formula V-2a are preferably selected from the group of the compounds of the formulae V-2a-1 and V-2a-2, preferably the compounds of the formula V-2a-1, more preferably these compounds of the formula V predominantly consist, even more preferably essentially consist and very particularly preferably completely consist thereof:

V-2a-1

$$\mathbb{R}^{51}$$
 \mathbb{R}^{52}

-continued

$$\mathbb{R}^{51}$$
 \mathbb{R}^{52}

10 in which

 R^{51} has the meaning indicated above and preferably denotes C_nH_{2n+1} or CH_2 —CH— $(CH_2)_Z$, and

 ${
m R}^{52}$ has the meaning indicated above and preferably denotes ${
m C}_m{
m H}_{2m+1}$ or ${
m O}-{
m C}_m{
m H}_{2m+1}$ or $({
m CH}_2)_Z$ — ${
m CH}={
m CH}_2$, and in which

n and m, independently of one another, denote an integer in the range from 0 to 15, preferably in the range from 1 to 7 and particularly preferably 1 to 5, and

z denotes 0, 1, 2, 3 or 4, preferably 0 or 2.

Preferred combinations of (R⁵¹ and R⁵²), in particular in the case of formula V-2a-1, are (C_nH_{2n+1} and C_mH_{2m+1}), (C_nH_{2n+1} and O— C_mH_{2m+i}), (CH_2 —CH—(CH_2) $_Z$ and C— C_mH_{2m+1}), (CH_2 —CH—(CH_2) $_Z$ and O— C_mH_{2m+1}) and (C_nH_{2n+1} and (CH_2) $_Z$ —CH— CH_2).

Preferred compounds of the formula V-2b are the compounds of the formula V-2b-1:

V-2b-1

V-2a-2

$$\mathbb{R}^{51} \longrightarrow \mathbb{R}^{52}$$

in which

 R^{51} has the meaning indicated above and preferably denotes C_nH_{2n+1} or CH_2 —CH— $(CH_2)_Z$, and

 R^{52} has the meaning indicated above and preferably denotes C_mH_{2m+1} or $O-C_mH_{2m+1}$ or $(CH_2)_Z$ — $CH-CH_2$, and in which

n and m, independently of one another, denote an integer in the range from 0 to 15, preferably in the range from 1 to 7 and particularly preferably 1 to 5, and

z denotes 0, 1, 2, 3 or 4, preferably 0 or 2.

The preferred combination of (R^{51} and R^{52}) here is, in particular, (C_nH_{2n+1} and C_mH_{2m+1}).

Preferred compounds of the formula V-2c are the compounds of the formula V-2c-1:

V-2c-1

$$\mathbb{R}^{51} \underbrace{\hspace{1.5cm}}^{F} \mathbb{R}^{52}$$

in which

 R^{51} has the meaning indicated above and preferably denotes C_nH_{2n+1} or CH_2 —CH— $(CH_2)_Z$, and

65 R⁵² has the meaning indicated above and preferably denotes C_mH_{2m+1} or $O-C_mH_{2m+1}$ or $(CH_2)_Z$ — $CH-CH_2$, and in which

n and m, independently of one another, denote an integer in the range from 0 to 15, preferably in the range from 1 to 7 and particularly preferably 1 to 5, and

z denotes 0, 1, 2, 3 or 4, preferably 0 or 2.

The preferred combination of $(R^{51} \text{ and } R^{52})$ here is, in ⁵ particular, $(C_nH_{2n+1} \text{ and } C_mH_{2m+1})$.

Preferred compounds of the formula V-2d are the compounds of the formula V-2d-1:

V-2d-1

$$\mathbb{R}^{51}$$
 \mathbb{R}^{52}

in which

 R^{51} has the meaning indicated above and preferably denotes C_nH_{2n+1} or CH_2 —CH— $(CH_2)_Z$, and

 R^{52} has the meaning indicated above and preferably denotes C_mH_{2m+1} or $O-C_mH_{2m+1}$ or $(CH_2)_Z$ —CH— CH_2 , and in which

n and m, independently of one another, denote an integer in the range from 0 to 15, preferably in the range from 1 to 7 and particularly preferably 1 to 5, and

z denotes 0, 1, 2, 3 or 4, preferably 0 or 2.

The preferred combination of (R^{51} and R^{52}) here is, in particular, (C_nH_{2n+1} and C_mH_{2m+1}).

Preferred compounds of the formula V-2e are the compounds of the formula V-2e-1:

 $R^{51} \xrightarrow{\qquad \qquad \qquad } R^{52}$

in which

 $\rm R^{51}$ has the meaning indicated above and preferably denotes $\rm C_{\it n}H_{\it 2n+1}$ or $\rm CH_2$ —CH—(CH₂)_{\it Z}, and

 R^{52} has the meaning indicated above and preferably denotes C_mH_{2m+1} or $O-C_mH_{2m+1}$ or $(CH_2)_Z$ — $CH-CH_2$, and in which

n and m, independently of one another, denote an integer in the range from 0 to 15, preferably in the range from 1 to 7 and particularly preferably 1 to 5, and

z denotes 0, 1, 2, 3 or 4, preferably 0 or 2.

The preferred combination of (R⁵¹ and R⁵²) here is, in particular, (C_nH_{2n+1} and $O-C_mH_{2m+1}$).

Preferred compounds of the formula V-2f are the compounds of the formula V-2f-1:

$$V-2f-1$$
 e

in which

 R^{51} has the meaning indicated above and preferably denotes C_nH_{2n+1} or CH_2 —CH— $(CH_2)_Z$, and

 R^{52} has the meaning indicated above and preferably denotes C_mH_{2m+1} or $O-C_mH_{2m+1}$ or $(CH_2)_Z$ —CH— CH_2 , and in which

n and m, independently of one another, denote an integer in the range from 0 to 15, preferably in the range from 1 to 7 and particularly preferably 1 to 5, and

z denotes 0, 1, 2, 3 or 4, preferably 0 or 2.

The preferred combinations of (R^{51} and R^{52}) here are, in particular, (C_nH_{2n+1} and C_mH_{2m+1}) and (C_nH_{2n+1} and O— C_mH_{2m+1}), particularly preferably (C_nH_{2n+1} and C_mH_{2m+1}).

Preferred compounds of the formula V-2g are the compounds of the formula V-2g-1:

 $\mathbb{R}^{51} - \mathbb{R}^{52}$

in which

 R^{51} has the meaning indicated above and preferably denotes C_nH_{2n+1} or CH_2 —CH— $(CH_2)_Z$, and

 ${
m R}^{52}$ has the meaning indicated above and preferably denotes ${
m C}_m{
m H}_{2m+1}$ or ${
m O}-{
m C}_m{
m H}_{2m+1}$ or $({
m CH}_2)_Z$ — ${
m CH}-{
m CH}_2$, and in which

n and m, independently of one another, denote an integer in the range from 0 to 15, preferably in the range from 1 to 7 and particularly preferably 1 to 5, and

z denotes 0, 1, 2, 3 or 4, preferably 0 or 2.

The preferred combinations of $(R^{51}$ and $R^{52})$ here are, in particular, $(C_nH_{2n+1}$ and $C_mH_{2m+1})$ and $(C_nH_{2n+1}$ and $O-C_mH_{2m+1})$, particularly preferably (C_nH_{2n+1}) and $O-C_mH_{2m+1}$.

The compounds of the formula VI are preferably selected from the group of the compounds of the formulae VI-1 to VI-4, more preferably these compounds of the formula VI predominantly consist, even more preferably essentially consist and very particularly preferably completely consist thereof:

$$R^{61} \longrightarrow A^{61} \longrightarrow A^{62} \longrightarrow C \equiv C \longrightarrow A^{63} \longrightarrow R^{62}$$

$$VI-2$$

$$R^{61} \longrightarrow A^{61} \longrightarrow C \equiv C \longrightarrow A^{62} \longrightarrow A^{63} \longrightarrow R^{62}$$

$$VI-3$$

$$R^{61} \longrightarrow A^{61} \longrightarrow A^{62} \longrightarrow Z^{62} \longrightarrow A^{63} \longrightarrow R^{62}$$

VI-4

-continued

 R^{61} A^{61} Z^{61} A^{62} A^{63} R^{62}

in which

 Z^{61} and Z^{62} denote trans-CH=CH— or trans-CF=CF—, preferably trans-CH=CH—, and the other parameters have the meaning given above under formula VI and preferably

 R^{61} and R^{62} , independently of one another, denote H, unfluorinated alkyl or alkoxy having 1 to 7 C atoms or unfluorinated alkenyl having 2 to 7 C atoms,

 X^{62} denotes F, Cl, —CN or —NCS, preferably —NCS, and one of

$$A^{61}$$
 to A^{63}

denotes

$$\begin{array}{c} F \\ \hline \\ F \\ \hline \\ \end{array}$$
 or
$$\begin{array}{c} F \\ \hline \\ \end{array}$$

and the others, independently of one another, denote

preferably

and preferably

 R^{61} denotes C_nH_{2n+1} or CH_2 =CH- $(CH_2)_Z$, and

 $\rm R^{62}$ denotes $\rm C_m H_{2m+1}$ or $\rm O-\!\!\!\!\!-C_m H_{2m+1}$ or $\rm (CH_2)_Z-\!\!\!\!\!-CH=\!\!\!\!\!\!\!-CH_2,$ and in which

n and m, independently of one another, denote an integer in the range from 0 to 15, preferably in the range from 1 to 7 and particularly preferably 1 to 5, and

z denotes 0, 1, 2, 3 or 4, preferably 0 or 2.

The compounds of the formula VI-1 are preferably selected from the group of the compounds of the formulae VI-1a and VI-1b, preferably selected from compounds of the formula VI-1a, more preferably these compounds of the formula VI predominantly consist, even more preferably essentially consist and very particularly preferably completely consist thereof:

 R^{61} C C F VI-1a VI-1a VI-1a

in which

35

40

55

 R^{61} has the meaning indicated above and preferably denotes C_nH_{2n+1} or CH_2 —CH— $(CH_2)_Z$, and

 ${\bf R}^{62}$ has the meaning indicated above and preferably denotes ${\bf C}_m{\bf H}_{2m+1}$ or ${\bf O}{--}{\bf C}_m{\bf H}_{2m+1}$ or $({\bf CH}_2)_Z{--}{\bf CH}{=-}{\bf CH}_2,$ and in which

n and m, independently of one another, denote an integer in the range from 0 to 15, preferably in the range from 1 to 7 and particularly preferably 1 to 5, and

z denotes 0, 1, 2, 3 or 4, preferably 0 or 2.

The preferred combinations of (R^{61} and R^{62}) here are, in particular, (C_nH_{2n+1} and C_mH_{2m+1}) and (C_nH_{2n+1} and O— C_mH_{2m+1}), in the case of formula VI-1a particularly preferably (C_nH_{2n+1} and C_mH_{2m+1}) and in the case of formula VI-1b particularly preferably (C_nH_{2n+1} and O— C_mH_{2m+1}).

The compounds of the formula VI-3 are preferably compounds of the formula VI-3a:

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45

55

$$R^{61} \xrightarrow{\text{VI-3a}} \text{T} = C \xrightarrow{\text{F}} X^{62}$$

in which the parameters have the meaning given above under formula VI-3 and preferably

 ${\bf R}^{61}$ has the meaning indicated above and preferably denotes ${\bf C}_n{\bf H}_{2n+1},$ in which

n denotes an integer in the range from 0 to 7, preferably in the 15 range from 1 to 5, and

X⁶² denotes —F, —Cl, —OCF₃, —CN or —NCS, particularly preferably —NCS.

The compounds of the formula VI-4 are preferably compounds of the formula VI-4-a:

$$R^{61} \xrightarrow{H} C \xrightarrow{H} C \xrightarrow{F} X^{62}$$

in which the parameters have the meaning given above under formula VI-4 and preferably

 R^{61} has the meaning indicated above and preferably denotes C_nH_{2n+1} , in which

n denotes an integer in the range from 0 to 7, preferably in the range from 1 to 5, and

X⁶² denotes F, Cl, OCF₃, —CN or —NCS, particularly preferably —NCS.

Further preferred compounds of the formula VI are the compounds of the following formulae:

$$C_nH_{2n+1}$$
 $C \equiv C$
 F
 C_nH_{2n+1}
 F
 $C \equiv C$
 F
 C_nH_{2n+1}
 F
 $C \equiv C$
 F
 C_nH_{2n+1}
 F
 $C \equiv C$
 $C \equiv$

in which

n denotes an integer in the range from 0 to 7, preferably in the range from 1 to 5.

The compounds of the formula VII are preferably selected from the group of the compounds of the formulae VII-1 to VII-6, more preferably these compounds of the formula VII predominantly consist, even more preferably essentially consist and very particularly preferably completely consist thereof:

$$X^{71}$$
 X^{72}
 Y^{71}
 Y^{72}
 Y^{71}

$$\mathbb{R}^{71} \underbrace{\hspace{1.5cm}}^{F} \underbrace{\hspace{1.5cm}}^{F} \mathbb{R}^{72}$$

$$R^{71}$$
 R^{72}
 $VII-4$

$$\mathbb{R}^{71} \underbrace{\hspace{1cm}}^{F} \underbrace{\hspace{1cm}}^{F} \underbrace{\hspace{1cm}}^{VII-5}$$

$$\mathbb{R}^{71} \underbrace{\hspace{1.5cm} \begin{array}{c} F \\ \\ \end{array}}_{\text{VII-6}} \underbrace{\hspace{1.5cm} \begin{array}{c} Y^{71} \\ \\ \end{array}}_{\text{VII-6}} \mathbb{R}^{72}$$

$$R^{71}$$
 R^{72} R^{72} R^{72}

where the compounds of the formula VII-5 are excluded from the compounds of the formula VII-6, and

50 in which the parameters have the respective meanings indicated above for formula VII, and preferably

R⁷¹ denotes unfluorinated alkyl or alkoxy, each having 1 to 7 C atoms, or unfluorinated alkenyl having 2 to 7 C atoms,

 R^{72} denotes unfluorinated alkyl or alkoxy, each having 1 to 7 $\,$ C atoms, or unfluorinated alkenyl having 2 to 7 C atoms, and

X⁷² denotes F, Cl or —OCF₃, preferably F, and particularly preferably

 R^{71} has the meaning indicated above and preferably denotes C_nH_{2n+1} or CH_2 —CH— $(CH_2)_Z$, and

 R^{72} has the meaning indicated above and preferably denotes C_mH_{2m+1} or $O-C_mH_{2m+1}$ or $(CH_2)_Z-CH-CH_2$, and in which

n and m, independently of one another, denote an integer in the range from 0 to 15, preferably in the range from 1 to 7 and particularly preferably 1 to 5, and z denotes 0, 1, 2, 3 or 4, preferably 0 or 2.

-continued

The compounds of the formula VIII-1 are preferably selected from the group of the compounds of the formulae VII-1a to VII-1d, more preferably these compounds of the formula VII-1 predominantly consist, even more preferably essentially consist and very particularly preferably completely consist thereof:

$$\mathbb{R}^{71}$$
 \mathbb{R}^{71}
 \mathbb{R}^{72}

$$CH_2$$
= CH - $(CH_2)_z$
 F
 X^{72}
 YII -1c 25

$$\mathbb{R}^{71} - \mathbb{Q} = \mathbb{Q} \times \mathbb{Q}$$

in which \boldsymbol{X}^{72} has the meaning given above for formula VII-2 and

 R^{71} has the meaning indicated above and preferably denotes C_nH_{2n+1} , in which

n denotes 1 to 7, preferably 2 to 6, particularly preferably 2, 3 or 5, and

z denotes 0, 1, 2, 3 or 4, preferably 0 or 2, and X^{72} preferably denotes F.

The compounds of the formula VIII-2 are preferably selected from the group of the compounds of the formulae VIII-2a and VII-2b, preferably of the formula VII-2a, more preferably these compounds of the formula VII-2 predominantly consist, even more preferably essentially consist and very particularly preferably completely consist thereof:

$$\mathbb{R}^{71}$$
 \mathbb{R}^{72}

$$\mathbb{R}^{71}$$
 \mathbb{R}^{72}
 \mathbb{R}^{72}

VII-1a 10 in which

VII-1b

 R^{71} has the meaning indicated above and preferably denotes C_nH_{2n+1} or CH_2 —CH— $(CH_2)_Z$, and

 R^{72} has the meaning indicated above and preferably denotes C_mH_{2m+1} or $O-C_mH_{2m+1}$ or $(CH_2)_Z$ —CH— CH_2 , and in which

n and m, independently of one another, denote an integer in the range from 0 to 15, preferably in the range from 1 to 7 and particularly preferably 1 to 5, and

z denotes 0, 1, 2, 3 or 4, preferably 0 or 2.

The preferred combinations of $(R^{71}$ and $R^{72})$ here are, in particular, (C_nH_{2n+1}) and (C_nH_{2n+1}) and (C_nH_{2n+1}) and (C_nH_{2n+1}) , particularly preferably (C_nH_{2n+1}) and (C_nH_{2n+1}) .

The compounds of the formula VIII-3 are preferably compounds of the formula VII-3a:

$$R^{71}$$
 F
 F
 F
 F
 F
 F
 F
 F
 F

in which

 R^{72} has the meaning indicated above and preferably denotes C_mH_{2m+1} or $O-C_mH_{2m+1}$ or $(CH_2)_Z-CH-CH_2$, and in which

n and m, independently of one another, denote an integer in the range from 0 to 15, preferably in the range from 1 to 7 and particularly preferably 1 to 5, and

z denotes 0, 1, 2, 3 or 4, preferably 0 or 2.

The preferred combinations of $(R^{71}$ and $R^{72})$ here are, in particular, (C_nH_{2n+1}) and (C_nH_{2n+1}) and (C_nH_{2n+1}) and (C_nH_{2n+1}) , particularly preferably (C_nH_{2n+1}) and (C_mH_{2n+1}) .

The compounds of the formula VIII-4 are preferably compounds of the formula VII-4-a:

$$R^{71}$$
 R^{72}
 R^{72}

in which

VII-2a 60

 R^{71} has the meaning indicated above and preferably denotes C_nH_{2n+1} or CH_2 —CH— $(CH_2)_Z$, and

 R^{72} has the meaning indicated above and preferably denotes $C_m H_{2m+1}$ or $O - C_m H_{2m+1}$ or $(CH_2)_Z - CH - CH_2$, and in which

30

40

n and m, independently of one another, denote an integer in the range from 0 to 15, preferably in the range from 1 to 7 and particularly preferably 1 to 5, and

z denotes 0, 1, 2, 3 or 4, preferably 0 or 2.

The preferred combinations of $(R^{71} \text{ and } R^{72})$ here are, in 5 particular, $(C_nH_{2n+1} \text{ and } C_mH_{2m+1})$ and $(C_nH_{2n+1} \text{ and }$ O— C_mH_{2m+1}), particularly preferably (C_nH_{2n+1}) and C_m

The compounds of the formula VIII-5 are preferably selected from the group of the compounds of the formulae VIII-5a and VII-5b, preferably of the formula VII-5a, more preferably these compounds of the formula VII-5 predominantly consist, even more preferably essentially consist and very particularly preferably completely consist thereof:

VII-5a
$$R^{71} \longrightarrow R^{72}$$

$$VII-5b$$

$$R^{71} \longrightarrow R^{72}$$

$$VII-7b$$

in which

R⁷¹ has the meaning indicated above and preferably denotes C_nH_{2n+1} or CH_2 =CH- $(CH_2)_Z$, and

R⁷² has the meaning indicated above and preferably denotes C_mH_{2m+1} or $O-C_mH_{2m+1}$ or $(CH_2)_Z$ —CH— CH_2 , and in

n and m, independently of one another, denote an integer in the range from 0 to 15, preferably in the range from 1 to 7 and particularly preferably 1 to 5, and

z denotes 0, 1, 2, 3 or 4, preferably 0 or 2.

The preferred combinations of (R⁷¹ and R⁷²) here are, in particular, $(C_nH_{2n+1} \text{ and } C_mH_{2m+1})$ and $(C_nH_{2n+1} \text{ and }$ O— C_mH_{2m+1}), particularly preferably (C_nH_{2n+1}) and C_m H_{2m+1}).

The compounds of the formula VIII-6 are preferably selected from the group of the compounds of the formulae VIII-6a and VII-6b, more preferably these compounds of the formula VII-6 predominantly consist, even more preferably essentially consist and very particularly preferably com- 50 one of pletely consist thereof:

$$R^{71} \longrightarrow F \qquad F \qquad F \qquad F$$

$$R^{72} \longrightarrow VII-6b \qquad 60$$

$$R^{71} \longrightarrow F \qquad F \qquad F \qquad F$$

$$R^{72} \longrightarrow R^{72} \qquad 65$$

in which

R⁷¹ has the meaning indicated above and preferably denotes C_nH_{2n+1} or CH_2 —CH— $(CH_2)_Z$, and

R⁷² has the meaning indicated above and preferably denotes C_mH_{2m+1} or $O-C_mH_{2m+1}$ or $(CH_2)_Z$ —CH= CH_2 , and in

n and m, independently of one another, denote an integer in the range from 0 to 15, preferably in the range from 1 to 7 and particularly preferably 1 to 5, and

z denotes 0, 1, 2, 3 or 4, preferably 0 or 2.

The preferred combinations of $(R^{71}$ and $R^{72})$ here are, in particular, $(C_nH_{2n+1} \text{ and } C_mH_{2m+1})$ and $(C_nH_{2n+1} \text{ and } O-C_mH_{2m+1})$, particularly preferably $(C_nH_{2n+1} \text{ and } C_m$

The liquid-crystalline media in accordance with the present application preferably comprise in total 0 to 40%, preferably 0 to 30% and particularly preferably 5 to 25%, of compounds of the formula VIII.

The compounds of the formula VIII are preferably selected from the group of the compounds of the formulae VIII-1 to VIII-3, more preferably these compounds of the formula VIII predominantly consist, even more preferably essentially consist and very particularly preferably completely consist thereof:

$$R^{81} \longrightarrow C = C \longrightarrow R^{82}$$

$$VIII-1$$

$$VIII-2$$

$$R^{81}$$
 $C \equiv C$ $VIII-3$

in which

 Y^{81} and Y^{82} denotes H and the other denotes H or F, and R⁸¹ has the meaning indicated above and preferably denotes C_nH_{2n+1} or CH_2 —CH— $(CH_2)_{\mathbb{Z}_2}$ and

R⁸² has the meaning indicated above and preferably denotes C_mH_{2m+1} or $O-C_mH_{2m+1}$ or $(CH_2)_Z$ — $CH=CH_2$, and in

n and m, independently of one another, denote an integer in the range from 0 to 15, preferably in the range from 1 to 7 and particularly preferably 1 to 5, and

VII-6b 60 z denotes 0, 1, 2, 3 or 4, preferably 0 or 2.

The preferred combinations of (R⁸¹ and R⁸²) here are, in particular, $(C_nH_{2n+1} \text{ and } C_mH_{2m+1})$ and $(C_nH_{2n+1} \text{ and } O-C_mH_{2m+i})$, particularly preferably $(C_nH_{2n+1} \text{ and } C_m$ H_{2m+1}).

The compounds of the formula VIII-1 are preferably selected from the group of the compounds of the formulae VIII-1a to VIII-1c, more preferably these compounds of the formula VIII-1 predominantly consist, even more preferably essentially consist and very particularly preferably completely consist thereof:

in which

 R^{81} has the meaning indicated above and preferably denotes C_nH_{2n+1} or CH_2 —CH— $(CH_2)_Z$, and

 R^{82} has the meaning indicated above and preferably denotes C_mH_{2m+1} or $O-C_mH_{2m+1}$ or $(CH_2)_Z-CH-CH_2$, and in which n and m, independently of one another, denote an 30 integer in the range from 0 to 15, preferably in the range from 1 to 7 and particularly preferably 1 to 5, and

z denotes 0, 1, 2, 3 or 4, preferably 0 or 2.

The preferred combinations of $(R^{81}$ and $R^{82})$ here are, in particular, (C_nH_{2n+1}) and (C_mH_{2n+1}) and (C_nH_{2n+1}) and (C_nH_{2n+1}) , particularly preferably (C_nH_{2n+1}) and (C_mH_{2n+1}) .

The compounds of the formula VIII-2 are preferably compounds of the formula VIII-2a:

VIII-2a $R^{81} \longrightarrow R^{82}$

in which

 R^{81} has the meaning indicated above and preferably denotes C_nH_{2n+1} or CH_2 —CH— $(CH_2)_Z$, and

 R^{82} has the meaning indicated above and preferably denotes C_mH_{2m+1} or $O-C_mH_{2m+1}$ or $(CH_2)_Z$ — $CH-CH_2$, and in which

n and m, independently of one another, denote an integer in the range from 0 to 15, preferably in the range from 1 to 7 and particularly preferably 1 to 5, and

z denotes 0, 1, 2, 3 or 4, preferably 0 or 2.

The preferred combinations of $(R^{81}$ and $R^{82})$ here are, in particular, (C_nH_{2n+1}) and (C_mH_{2m+1}) , (C_nH_{2n+1}) and $(CH_2 = CH - (CH_2)_Z)$ and $(CH_2 = CH - (CH_2)_Z)$ and $(CH_2 = CH - (CH_2)_Z)$ and (C_mH_2) , particularly preferably (C_mH_2) and (C_mH_2) .

The compounds of the formula VIII-3 are preferably compounds of the formula VIII-3a:

VIII-3a

$$\mathbb{R}^{81} - \mathbb{C} = \mathbb{C} - \mathbb{R}^{82}$$

10 in which

 R^{81} has the meaning indicated above and preferably denotes C_nH_{2n+1} or CH_2 —CH— $(CH_2)_Z$, and

 ${
m R}^{82}$ has the meaning indicated above and preferably denotes ${
m C}_m{
m H}_{2m+1}$ or ${
m O}-{
m C}_m{
m H}_{2m+1}$ or $({
m CH}_2)_Z$ — ${
m CH}={
m CH}_2$, and in which

n and m, independently of one another, denote an integer in the range from 0 to 15, preferably in the range from 1 to 7 and particularly preferably 1 to 5, and

z denotes 0, 1, 2, 3 or 4, preferably 0 or 2.

The preferred combinations of $(R^{81}$ and $R^{82})$ here are, in particular, $(C_nH_{2n+1} \text{ and } C_mH_{2m+1})$ and $(C_nH_{2n+1} \text{ and } O-C_mH_{2m+1})$.

The compounds of the formula IX are preferably selected from the group of the compounds of the formulae IX-1 to IX-3, more preferably these compounds of the formula IX predominantly consist, even more preferably essentially consist and very particularly preferably completely consist thereof:

in which the parameters have the respective meaning indicated above under formula IX and preferably one of

$$A^{92}$$
 to A^{94}

denotes

IX-1e

-continued

and

in which

R⁹¹ has the meaning indicated above and preferably denotes C_nH_{2n+1} or CH_2 —CH— $(CH_2)_Z$, and

R⁹² has the meaning indicated above and preferably denotes C_mH_{2m+1} or $O-C_mH_{2m+1}$ or $(CH_2)_Z$ — $CH=CH_2$, and in 15 which

n and m, independently of one another, denote an integer in the range from 0 to 15, preferably in the range from 1 to 7 and particularly preferably 1 to 5, and

z denotes 0, 1, 2, 3 or 4, preferably 0 or 2.

The preferred combinations of (R⁹¹ and R⁹²) here are, in particular, (C_nH_{2n+1}) and C_mH_{2m+1} and (C_nH_{2n+1}) and $O - C_m H_{2m+1}$).

The liquid-crystalline media in accordance with the present application preferably comprise in total 5 to 30%, preferably 10 to 25% and particularly preferably 15 to 20%, of compounds of the formula IX.

The compounds of the formula IX-1 are preferably 30 selected from the group of the compounds of the formulae IX-1a to IX-1e, more preferably these compounds of the formula IX-1 predominantly consist, even more preferably essentially consist and very particularly preferably completely consist thereof:

$$\mathbb{R}^{91}$$
 \mathbb{R}^{91}
 \mathbb{R}^{92}
 $\mathbb{I}X-1b$ 45

$$R^{91}$$
 X^{92}

$$\mathbb{R}^{91}$$
 X^{92}

-continued

$$\mathbb{R}^{91} \underbrace{\hspace{1cm}}^{O} \underbrace{\hspace{1cm}}^{F} X^{92}$$

in which the parameters have the meaning given above and preferably

 $\ensuremath{R^{91}}$ has the meaning indicated above and preferably denotes C_nH_{2n+1} , and

n denotes an integer in the range from 0 to 15, preferably in the range from 1 to 7 and particularly preferably 1 to 5, and X⁹² preferably denotes F or Cl.

The compounds of the formula IX-2 are preferably selected from the group of the compounds of the formulae IX-2a and IX-2b, more preferably these compounds of the formula IX-2 predominantly consist, even more preferably essentially consist and very particularly preferably completely consist thereof:

$$R^{91}$$
 R^{92}
 $IX-2a$
 $IX-2b$

$$R^{91}$$
 R^{92}

in which

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IX-1c

R⁹¹ has the meaning indicated above and preferably denotes C_nH_{2n+1} or CH_2 =CH- $(CH_2)_Z$, and

R⁹² has the meaning indicated above and preferably denotes C_mH_{2m+1} or $O-C_mH_{2m+1}$ or $(CH_2)_Z$ — $CH-CH_2$, and in

50 n and m, independently of one another, denote an integer in the range from 0 to 15, preferably in the range from 1 to 7 and particularly preferably 1 to 5, and

z denotes 0, 1, 2, 3 or 4, preferably 0 or 2.

The preferred combination of (R91 and R92) here is, in particular, $(C_nH_{2n+1} \text{ and } C_mH_{2m+1})$.

The compounds of the formula IX-3 are preferably compounds of the formulae IX-3a and IX-3b:

$$R^{91} - X^{92}$$

IX-3h

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-continued

$$\mathbb{R}^{91}$$
 \longrightarrow $\mathbb{C}_m\mathbb{H}_{2m+1}$

in which

 R^{91} has the meaning indicated above and preferably denotes 10 C_nH_{2n+1} or CH_2 —CH— $(CH_2)_Z$, and

 R^{92} has the meaning indicated above and preferably denotes C_mH_{2m+1} or $O-C_mH_{2m+1}$ or $(CH_2)_Z$ — $CH-CH_2$, and in which

n and m, independently of one another, denote an integer in the range from 0 to 15, preferably in the range from 1 to 7 and particularly preferably 1 to 5, and

z denotes 0, 1, 2, 3 or 4, preferably 0 or 2.

The preferred combinations of $(R^{91}$ and $R^{92})$ here are, in particular, $(C_nH_{2n+1} \text{ and } C_mH_{2m+1})$ and $(C_nH_{2n+1} \text{ and } O-C_mH_{2n+1})$, particularly preferably $(C_nH_{2n+1} \text{ and } O-C_mH_{2m+1})$.

In a preferred embodiment of the present invention, the medium comprises one or more dielectrically positive compounds of the formula V-1 having a dielectric anisotropy of greater than 3.

The liquid-crystalline media in accordance with the present invention preferably comprise 10% or less, preferably 5% or less, particularly preferably 2% or less, very particularly preferably 1% or less, and in particular absolutely no compound having only two or fewer five- and/or six-membered rings.

In a preferred embodiment of the present invention, the medium comprises one or more compounds of the formula 35 VI.

In a further preferred embodiment of the present invention, the medium comprises one or more compounds of the formula $\overline{\rm VII}$.

The definitions of the abbreviations (acronyms) used for 40 the compounds in the present application are indicated below in Table D or are evident from Tables A to C.

In a preferred embodiment of the present invention, the liquid-crystal medium, or component A of the liquid-crystal medium, comprises one or more compounds of the formulae 45 IB-1 and/or IB-2 and/or IB-3 and/or IB-4.

The liquid-crystal medium, or component A of the liquid-crystal medium, preferably comprises one or more compounds selected from the compounds of the formulae IB-1a-1 to IB-1a-12, particularly preferably of the formula IB-1a-2, 50 very particularly preferably one or more compounds of the formula IB-1a-2 and one or more compounds selected from the group of the compounds of the formula IB-1a-1 and formulae IB-1a-3 to IB-1a-12, and one or more compounds of the formulae IB-1b-1 to IB-1b-12 and/or IB-2 and/or IB-3 55 and/or IB-4.

In a further preferred embodiment of the present invention, the liquid-crystal medium, or component A of the liquid-crystal medium, comprises one or more compounds selected from the group of the compounds of the formulae IB-1b-1 to 60 IB-1b-12, particularly preferably selected from the group of the compounds of the formulae IB-1b-5 and/or IB-1b-7 and/or IB-1b-8 and/or IB-1b-9 and/or IB-1b-10, and one or more compounds selected from the group of the compounds of the formulae IB-1a-1 to IB-1a-12, preferably of the formulae IB-1a-2, and/or one or more compounds of the formulae IB-2 and/or IB-3 and/or IB-4.

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In a further preferred embodiment of the present invention, the liquid-crystal medium, or component A of the liquid-crystal medium, comprises one or more compounds of the formula IB-2 and one or more compounds of the formula IB-1, preferably of the formula IB-1a, preferably of the formulae IB-1a-2, and/or IB-1b, and/or one or more compounds of the formulae IB-3 and/or IB-4.

In a further preferred embodiment of the present invention, the liquid-crystal medium, or component A of the liquid-crystal medium, comprises one or more compounds of the formula IB-3 and one or more compounds of the formula IB-1, preferably of the formula IB-1a-2, and/or IB-1b, and/or one or more compounds of the formulae IB-2 and/or IB-4.

In a further preferred embodiment of the present invention, the liquid-crystal medium, or component A of the liquid-crystal medium, comprises one or more compounds of the formula IB-4 and one or more compounds of the formula IB-1, preferably of the formula IB-1a-2, and/or IB-1b, and/or one or more compounds of the formulae IB-2 and/or IB-3.

The liquid-crystalline media in accordance with the present invention preferably comprise, more preferably predominantly consist of, even more preferably essentially consist of and very particularly preferably completely consist of compounds selected from the group of the compounds of the formulae IA, IB, II, IV and V, preferably IA, IB, IIA, IIB and IV, or selected from the group of the compounds of the formulae IA, IB, IIIA, IIIB, IV and V, preferably IA, IB, MA, IIIB, and IV

In this application, comprise in connection with compositions means that the entity in question, i.e. the medium or the component, comprises the component or components or compound or compounds indicated, preferably in a total concentration of 10% or more and very preferably 20% or more.

In this connection, predominantly consist of means that the entity in question comprises 55% or more, preferably 60% or more and very preferably 70% or more, of the component or components or compound or compounds indicated.

In this connection, essentially consist of means that the entity in question comprises 80% or more, preferably 90% or more and very preferably 95% or more, of the component or components or compound or compounds indicated.

In this connection, completely consist of means that the entity in question comprises 98% or more, preferably 99% or more and very preferably 100.0% of the component or components or compound or compounds indicated.

Other mesogenic compounds which are not explicitly mentioned above can optionally and advantageously also be used in the media in accordance with the present invention. Such compounds are known to the person skilled in the art.

The liquid-crystal media in accordance with the present invention preferably have a clearing point of 90° C. or more, more preferably 100° C. or more, even more preferably 120° C. or more, particularly preferably 150° C. or more and very particularly preferably 170° C. or more.

The nematic phase of the media according to the invention preferably extends at least from 20° C. or less to 90° C. or more, preferably up to 100° C. or more, more preferably at least from 0° C. or less to 120° C. or more, very preferably at least from -10° C. or less to 140° C. or more and in particular at least from -20° C. or less to 150° C. or more.

The $\Delta \varepsilon$ of the liquid-crystal medium in accordance with the invention, at 1 kHz and 20° C., is preferably 1 or more, more preferably 2 or more and very preferably 3 or more.

The Δn of the liquid-crystal media in accordance with the present invention, at 589 nm (Na^D) and 20° C., is preferably

in the range from 0.200 or more to 0.90 or less, more preferably in the range from 0.250 or more to 0.90 or less, even more preferably in the range from 0.300 or more to 0.85 or less and very particularly preferably in the range from 0.350 or more to 0.800 or less.

In a preferred embodiment of the present application, the Δn of the liquid-crystal media in accordance with the present invention is preferably 0.40 or more, more preferably 0.45 or more.

In accordance with the present invention, the individual 10 compounds of the formula IA in the liquid-crystal media are preferably used in a total concentration of 10% to 100%, more preferably 30% to 95%, even more preferably 40% to 90% and very preferably 50% to 90%, of the mixture as a whole.

In accordance with the present invention, the individual compounds of the formula IB in the liquid-crystal media are preferably used in a total concentration of 10% to 100%, more preferably 30% to 95%, even more preferably 40% to 90% and very preferably 50% to 90%, of the mixture as a whole.

In the embodiment of the present invention in which the 20 liquid-crystal media comprise one or more compounds selected from the group of the compounds of the formulae IIA and IIB, the further compounds are preferably employed as follows.

The compounds selected from the group of the compounds 25 of the formulae IIA and IIB are preferably used in a total concentration of 1% to 30%, more preferably 2% to 20%, even more preferably 3% to 18% and very preferably 4% to 16%, of the mixture as a whole.

The compounds of the formula IV are preferably used in a 30 total concentration of 1% to 20%, more preferably 2% to 15%, even more preferably 3% to 12% and very preferably 5% to 10%, of the mixture as a whole.

The liquid-crystal media preferably comprise, more preferably predominantly consist of and very preferably completely consist of in total 70% to 100%, more preferably 80% to 100% and very preferably 90% to 100% and in particular 95% to 100%, of the compounds of the formulae IA, IB, IIA, IIB and IV to IX, preferably of the formulae I, IIA, IIB and IV.

In the embodiment of the present invention in which the 40 liquid-crystal media comprise one or more compounds selected from the group of the compounds of the formulae IIIA and IIIB, the further compounds are preferably employed as follows.

The compounds selected from the group of the compounds 45 of the formulae IIIA and IIIB are preferably used in a total concentration of 1% to 60%, more preferably 5% to 55%, even more preferably 7% to 50% and very preferably 10% to 45%, of the mixture as a whole.

If the liquid-crystal media comprise only one or more 50 compounds of the formula IIIA, but no compounds of the formula IIIB, the compounds of the formula IIIA are preferably used in a total concentration of 10% to 60%, more preferably 20% to 55%, even more preferably 30% to 50% and very preferably 35% to 45%, of the mixture as a whole. 55

If the liquid-crystal media comprise only one or more compounds of the formula MB, but no compounds of the formula IIIA, the compounds of the formula IIIB are preferably used in a total concentration of 5% to 45%, more preferably 10% to 40%, even more preferably 15% to 35% and 60 very preferably 20% to 30%, of the mixture as a whole.

If the liquid-crystal media comprise both one or more compounds of the formula IIIA and one or more compounds of the formula IIIB, the compounds of the formula IIIA are preferably used in a total concentration of 5% to 50%, more 65 preferably 10% to 45%, even more preferably 15% to 30% and very preferably 20% to 25%, of the mixture as a whole

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and the compounds of the formula IIIB are preferably used in a total concentration of 1% to 35%, more preferably 5% to 30%, even more preferably 7% to 25% and very preferably 10% to 20%, of the mixture as a whole.

The compounds of the formula IV are preferably used in a total concentration of 1% to 20%, more preferably 2% to 15%, even more preferably 3% to 12% and very preferably 5% to 10%, of the mixture as a whole.

The liquid-crystal media preferably comprise, more preferably predominantly consist of and very preferably completely consist of in total 70% to 100%, more preferably 80% to 100% and very preferably 90% to 100% and in particular 95% to 100%, of compounds selected from the group of the compounds of the formulae IA, IB, IIIA, IIIB and IV to IX, preferably of the formulae IA, IB and/or IIIA.

In a particularly preferred embodiment of the present invention, the liquid-crystalline media comprise one or more compounds of the formula V and one or more compounds of the formula VI.

In a further particularly preferred embodiment of the present invention, the liquid-crystalline media comprise one or more compounds of the formula V and one or more compounds of the formula VII.

The liquid-crystalline media in accordance with the present invention likewise preferably comprise one or more compounds of the formula V, one or more compounds of the formula VI and one or more compounds of the formula VIII.

If the liquid-crystalline media in accordance with the present application comprise one or more compounds of the formula V, the concentration of these compounds is preferably in total 10 to 30%, preferably 15 to 25% and particularly preferably 18 to 22%.

If the liquid-crystalline media in accordance with the present application comprise one or more compounds of the formula VI, the concentration of these compounds is preferably in total 15 to 35%, preferably 18 to 30% and particularly preferably 22 to 26%.

If the liquid-crystalline media in accordance with the present application comprise one or more compounds of the formula VII, the concentration of these compounds is preferably in total 4 to 25%, preferably 8 to 20% and particularly preferably 10 to 14%.

If the liquid-crystalline media in accordance with the present application comprise one or more compounds of the formula VIII, the concentration of these compounds is preferably in total 15 to 35%, preferably 18 to 30% and particularly preferably 22 to 26%.

If the liquid-crystalline media in accordance with the present application comprise one or more compounds of the formula IX, the concentration of these compounds is preferably in total 5 to 25%, preferably 10 to 20% and particularly preferably 13 to 17%.

In the present application, the expression dielectrically positive describes compounds or components where $\Delta \epsilon > 3.0$, dielectrically neutral describes those where $-1.5 \le \Delta \epsilon \le 3.0$ and dielectrically negative describes those where $\Delta \epsilon < -1.5$. $\Delta \epsilon$ is determined at a frequency of 1 kHz and at 20° C. The dielectric anisotropy of the respective compound is determined from the results of a solution of 10% of the respective individual compound in a nematic host mixture. If the solubility of the respective compound in the host mixture is less than 10%, the concentration is reduced to 5%. The capacitances of the test mixtures are determined both in a cell having homeotropic alignment and in a cell having homogeneous alignment. The cell thickness of both types of cells is approximately 20 μ m. The voltage applied is a rectangular wave having a frequency of 1 kHz and an effective value of typi-

cally 0.5 V to 1.0 V, but it is always selected to be below the capacitive threshold of the respective test mixture.

The following definitions apply here.

$$\Delta \epsilon = (\epsilon_{\parallel} - \epsilon_{\perp})$$
 and

$$\epsilon_{average} = (\epsilon_{\parallel} + 2\epsilon_{\perp})/3.$$

The host mixture used for dielectrically positive compounds is mixture ZLI-4792 and that used for dielectrically neutral and dielectrically negative compounds is mixture 10 ZLI-3086, both from Merck KGaA, Germany. The absolute values of the dielectric constants of the compounds are determined from the change in the respective values of the host mixture on addition of the compounds of interest. The values are extrapolated to a concentration of the compounds of interest of 100%.

Components having a nematic phase at the measurement temperature of 20° C. are measured as such, all others are treated like compounds.

The expression threshold voltage in the present application 20 refers to the optical threshold and is quoted for 10% relative contrast (V_{10}) , and the expression saturation voltage refers to the optical saturation and is quoted for 90% relative contrast (V_{90}) , in both cases unless expressly stated otherwise. The capacitive threshold voltage (V_0) , also called the Freedericks 25 threshold (V_{Fe}) , is only used if expressly mentioned.

The parameter ranges indicated in this application all include the limit values, unless expressly stated otherwise.

The different upper and lower limit values indicated for various ranges of properties in combination with one another 30 give rise to additional preferred ranges.

Throughout this application, the following conditions and definitions apply, unless expressly stated otherwise. All concentrations are quoted in percent by weight and relate to the respective mixture as a whole, all temperatures are quoted in 35 degrees Celsius and all temperature differences are quoted in differential degrees. All physical properties are determined in accordance with "Merck Liquid Crystals, Physical Properties of Liquid Crystals", Status November 1997, Merck KGaA, Germany, and are quoted for a temperature of 20° C., unless 40 expressly stated otherwise. The optical anisotropy (Δn) is determined at a wavelength of 589.3 nm. The dielectric anisotropy ($\Delta \epsilon$) is determined at a frequency of 1 kHz. The threshold voltages, as well as all other electro-optical properties, are determined using test cells produced at Merck 45 KGaA, Germany. The test cells for the determination of $\Delta \epsilon$ have a cell thickness of approximately 20 µm. The electrode is a circular ITO electrode having an area of 1.13 cm² and a guard ring. The orientation layers are SE-1211 from Nissan Chemicals, Japan, for homeotropic orientation (ϵ_{\parallel}) and polyimide AL-1054 from Japan Synthetic Rubber, Japan, for homogeneous orientation (ϵ_1) . The capacitances are determined using a Solatron 1260 frequency response analyser using a sine wave with a voltage of $0.3 V_{rms}$.

The light used in the electro-optical measurements is white 55 light. A set-up using a commercially available DMS instrument from Autronic-Melchers, Germany, is used here. The characteristic voltages have been determined under perpendicular observation. The threshold (V_{10}) , mid-grey (V_{50}) and saturation (V_{90}) voltages have been determined for 10%, 50% 60 and 90% relative contrast, respectively.

The liquid-crystalline media are investigated with respect to their properties in the microwave frequency region as described in A. Penirschke et al., "Cavity Perturbation Method for Characterisation of Liquid Crystals up to 35 GHz", 34th European Microwave Conference—Amsterdam, pp. 545-548. Compare in this respect also A. Gaebler et al.,

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"Direct Simulation of Material Permittivites . . . ", 12MTC 2009—International Instrumentation and Measurement Technology Conference, Singapore, 2009 (IEEE), pp. 463-467, and DE 10 2004 029 429 A, in which a measurement method is likewise described in detail.

The liquid crystal is introduced into a cylindrical polytetrafluoroethylene (PTFE) or quartz capillary. The capillary has an internal radius of 180 µm and an external radius of 350 µm. The effective length is 2.0 cm. The filled capillary is introduced into the centre of the cylindrical cavity with a resonance frequency of 19 GHz. This cavity has a length of 11.5 mm and a radius of 6 mm. The input signal (source) is then applied, and the result of the output signal is recorded using a commercial vector network analyser. For other frequencies, the dimensions of the cavity are adapted correspondingly.

The change in the resonance frequency and the Q factor between the measurement with the capillary filled with the liquid crystal and the measurement without the capillary filled with the liquid crystal is used to determine the dielectric constant and the loss angle at the corresponding target frequency by means of equations 10 and 11 of the above-mentioned publication A. Penirschke et al., "Cavity Perturbation Method for Characterisation of Liquid Crystals up to 35 GHz", 34th European Microwave Conference—Amsterdam, pp. 545-548, as described therein.

The values for the components of the properties perpendicular and parallel to the director of the liquid crystal are obtained by alignment of the liquid crystal in a magnetic field. To this end, the magnetic field of a permanent magnet is used. The strength of the magnetic field is 0.35 tesla. The alignment of the magnet is set correspondingly and then rotated correspondingly through 90°.

Preferred components are phase shifters, varactors, wireless and radio wave antenna arrays, matching circuit adaptive filters and others.

In the present application, the term compounds is taken to mean both one compound and a plurality of compounds, unless expressly stated otherwise.

The liquid-crystal media according to the invention preferably have nematic phases of in each case at least from -20° C. to 80° C., preferably from -30° C. to 85° C. and very particularly preferably from -40° C. to 100° C. The phase particularly preferably extends to 120° C. or more, preferably to 140° C. or more and very particularly preferably to 160° C. or more. The expression have a nematic phase here means on the one hand that no smectic phase and no crystallisation are observed at low temperatures at the corresponding temperature and on the other hand that no clearing occurs on heating from the nematic phase. The investigation at low temperatures is carried out in a flow viscometer at the corresponding temperature and checked by storage in test cells having a layer thickness of 5 µm for at least 100 hours. At high temperatures, the clearing point is measured in capillaries by conventional methods.

Furthermore, the liquid-crystal media according to the invention are characterised by high optical anisotropies in the visible region. The birefringence at 589 nm is preferably 0.20 or more, particularly preferably 0.25 or more, particularly preferably 0.40 or more and very particularly preferably 0.45 or more. In addition, the birefringence is preferably 0.80 or less.

In a preferred embodiment of the present invention, the liquid-crystal media employed have positive dielectric anisotropy (). This is preferably 1.8 or more and 15.0 or less, more preferably between 2.0 or more and 10.0 or less, par-

ticularly preferably between 3.0 or more and 8.0 or less and very particularly preferably between 3.5 or more and 6.0 or less.

If the liquid-crystal media employed have negative dielectric anisotropy (), this is preferably less than or equal to -2.5, particularly preferably less than or equal to -4.0 and very particularly preferably less than or equal to -5.0.

In this preferred embodiment of the present invention, in which the liquid-crystal media employed have negative dielectric anisotropy (), the value thereof is preferably between 1.5 or more and 15.0 or less, particularly preferably between 1.8 or more and 12.0 or less and very particularly preferably between 2.0 or more and 10.0 or less.

Furthermore, the liquid-crystal media according to the invention are characterised by high anisotropies in the microwave region and/or millimeter wave region. The birefringence is, for example, preferably 0.14 or more, particularly preferably 0.15 or more, particularly preferably 0.20 or more, particularly preferably 0.25 or more and very particularly preferably 0.30 or more, at about 8.3 GHz. In addition, the 20 birefringence is preferably 0.80 or less.

The dielectric anisotropy in the microwave region is defined as

 $\Delta \epsilon_r = (\epsilon_{r,\parallel} - \epsilon_{r,\perp}).$

The tuneability (τ) is defined as

 $\tau = (\Delta \epsilon_r / \epsilon_{r,\parallel}).$

The material quality (η) is defined as

 $\eta = (\tau/\tan \delta_{er,max.})$, where

the maximum dielectric loss is

$$\tan \delta_{\epsilon r, max.} = \max. \{ \tan \delta_{\epsilon r, \perp}; \tan \delta_{\epsilon r, \parallel} \}.$$

The material quality (η) of the preferred liquid-crystal 35 materials is 6 or more, preferably 8 or more, preferably 10 or more, preferably 15 or more, preferably 17 or more, preferably 20 or more, particularly preferably 25 or more, very particularly preferably 30 and in particular 40 or more or even 50 or more.

In the corresponding components, the preferred liquid-crystal materials have phase shifter qualities of 15° /dB or more, preferably 20° /dB or more, preferably 30° /dB or more, preferably 40° /dB or more, preferably 50° /dB or more, particularly preferably 80° /dB or more and very particularly 45 preferably 100° /dB or more.

In some embodiments, however, it is also possible to use liquid crystals having a negative value of the dielectric anisotropy.

The liquid crystals employed are either individual substances or mixtures. They preferably have a nematic phase.

The term "alkyl" preferably encompasses straight-chain and branched alkyl groups, as well as cycloalkyl groups, each having 1 to 15 carbon atoms, in particular the straight-chain groups methyl, ethyl, propyl, butyl, pentyl, hexyl and heptyl, 55 as well as cyclopropyl and cyclohexyl. Groups having 2 to 10 carbon atoms are generally preferred.

The term "alkenyl" preferably encompasses straight-chain and branched alkenyl groups having 2 to 15 carbon atoms, in particular the straight-chain groups. Particularly preferred 60 alkenyl groups are C_2 - to C_7 -1E-alkenyl, C_4 - to C_7 -3E-alkenyl, C_5 - to C_7 -4-alkenyl, C_6 - to C_7 -5-alkenyl and C_7 -6-alkenyl, in particular C_2 - to C_7 -1E-alkenyl, C_4 - to C_7 -3E-alkenyl and C_5 - to C_7 -4-alkenyl. Examples of further preferred alkenyl groups are vinyl, 1E-propenyl, 1E-butenyl, 65 1E-pentenyl, 1E-hexenyl, 1E-heptenyl, 3-butenyl, 3E-pentenyl, 3E-hexenyl, 3E-hexenyl, 4-pentenyl, 4Z-hexenyl,

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4E-hexenyl, 4Z-heptenyl, 5-hexenyl, 6-heptenyl and the like. Groups having up to 5 carbon atoms are generally preferred.

The term "fluoroalkyl" preferably encompasses straightchain groups having a terminal fluorine, i.e. fluoromethyl, 2-fluoroethyl, 3-fluoropropyl, 4-fluorobutyl, 5-fluoropentyl, 6-fluorohexyl and 7-fluoroheptyl. However, other positions of the fluorine are not excluded.

The term "oxaalkyl" or "alkoxyalkyl" preferably encompasses straight-chain radicals of the formula C_nH_{2n+1} —O— $(CH_2)_m$, in which n and m each, independently of one another, denote an integer from 1 to 10. Preferably, n here is 1 and m is 1 to 6.

Compounds containing a vinyl end group and compounds containing a methyl end group have low rotational viscosity.

In the present application, both high-frequency technology and hyperfrequency technology denote applications having frequencies in the range from 1 MHz to 100 THz, preferably from 1 GHz to 30 THz, more preferably 2 GHz to 10 THz, particularly preferably from about 5 GHz to 5 THz.

The liquid-crystal media in accordance with the present invention may comprise further additives and chiral dopants in the usual concentrations. The total concentration of these further constituents is in the range from 0% to 10%, preferably 0.1% to 6%, based on the mixture as a whole. The concentrations of the individual compounds used are each preferably in the range from 0.1% to 3%. The concentration of these and similar additives is not taken into consideration when quoting the values and concentration ranges of the liquid-crystal components and liquid-crystal compounds of the liquid-crystal media in this application.

The liquid-crystal media according to the invention consist of a plurality of compounds, preferably 3 to 30, more preferably 4 to 20 and very preferably 4 to 15 compounds. These compounds are mixed in a conventional manner. In general, the desired amount of the compound used in the smaller amount is dissolved in the compound used in the larger amount. If the temperature is above the clearing point of the compound used in the higher concentration, it is particularly easy to observe completion of the dissolution process. It is, however, also possible to prepare the media in other conventional ways, for example using so-called pre-mixes, which can be, for example, homologous or eutectic mixtures of compounds, or using so-called "multibottle" systems, the constituents of which are themselves ready-to-use mixtures.

All temperatures, such as, for example, the melting point T(C,N) or T(C,S), the transition from the smectic (S) to the nematic (N) phase T(S,N) and the clearing point T(N,I) of the liquid crystals, are quoted in degrees Celsius. All temperature differences are quoted in differential degrees.

In the present invention and especially in the following examples, the structures of the mesogenic compounds are indicated by means of abbreviations, also referred to as acronyms. In these acronyms, the chemical formulae are abbreviated as follows using Tables A to C below. All groups $C_nH_{2n+1},\ C_mH_{2m+1}$ and C_lH_{2l+1} or $C_nH_{2n-1},\ C_mH_{2m-1}$ and C_lH_{2l-1} denote straight-chain alkyl or alkenyl, preferably 1E-alkenyl, having n, m and 1C atoms respectively, where n, m and 1, independently of one another, denote an integer from 1 to 9, preferably 1 to 7, or from 2 to 9, preferably 2 to 7, respectively. C_oH_{2o+1} denotes straight-chain alkyl having 1 to 7, preferably 1 to 4, C atoms, or branched alkyl having 1 to 7, preferably 1 to 4, C atoms.

Table A lists the codes used for the ring elements of the core structures of the compounds, while Table B shows the linking groups. Table C gives the meanings of the codes for the left-hand or right-hand end groups. Table D shows illustrative structures of compounds with their respective abbreviations.

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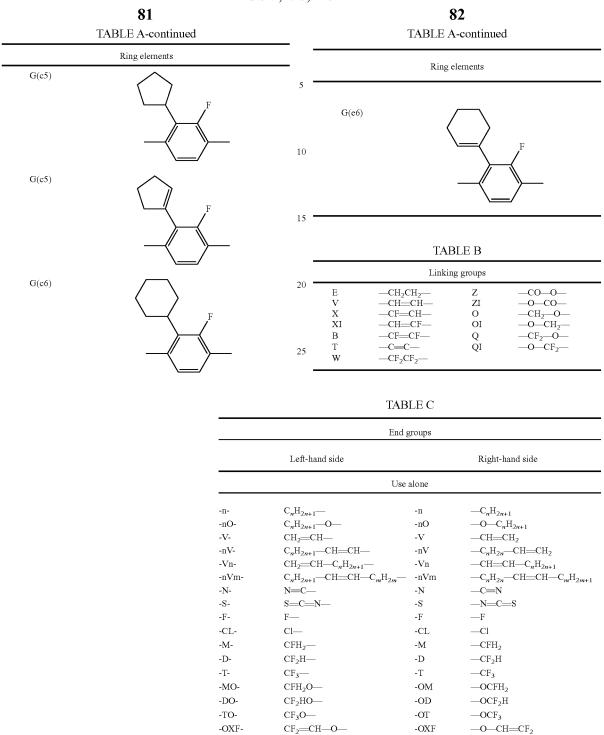
	Ring elements			Ring elements
С	<u> </u>	5	N(1, 4)	
D		10	N3f	F F
A		15		
G	F	20	tH	
U	F F	25	tH2f	F
Y	F F	30	K	F F F
fX	- F	35	L	
	F F O F	40	F	F
M	$-\!\!\!\!\!\!\!\!\!\!\!\!\!\!\!\!\!\!\!\!\!\!\!\!\!\!\!\!\!\!\!\!\!\!\!\!$	45	P(o)	C _o H _{2o+1}
N	N	50	P(i3)	
fN	~~~	55	2 (40)	
dH	F	60	P(t4)	
		65		

	Ring elements			Ring elements
P(c3)		5	P(3xo)	$o \in \{1; 2; 3; 4; 5; 6\}$ $H(CH_2)_o$ $(CH_2)_oH$
P(c4)		10		(CH ₂) _o H
		15	P(4xo)	$o \in \{1; 2; 3; 4; 5; 6\}$ $H(CH_2)_o$ (CH ₂) _o H
P(c5)		20		
		20		$H(CH_2)_{o}'$ $(CH_2)_{o}H$ $o \in \{1; 2; 3; 4; 5; 6\}$
P(e5)		25	GI(o)	F (CH ₂) ₀ H
		30	GI(i3)	$o \in \{1; 2; 3; 4; 5; 6\}$
P(c6)		35		F
			GI(t4)	F. V
P(e6)		40		
- ()		45	GI(c3)	F,
		50		
P(0, 5F)	$(CH_2)_{o}H$		GI(c4)	F,
	F	55		
P(0, 6F)	o ∈ {1, 2, 3, 4, 5, 6} (CH ₂) _o H	60	GI(c5)	
	F	65		\ <u></u> /

	Ring elements	-		Ring elements
GI(e5)	F,	5	NI	~\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\
		10	fNI	N_N_
GI(c6)	F,	15	N(2, 6)	F
		20	N3fI	F, F
GI(e6)	F __	20		F
		25	tHI	
P		30	tH2fI	F F
DI		35		
AI		40	KI	F F F
GI	F	45	LI	
UI	F		FI	F _v
	F	50		
fXI	F O F	55	PI(o)	C _o H _{2o+1}
MI		60	PI(c3)	
1411	N=	65		

80 TABLE A-continued

	17 IDEE 77 continued		- MBBBA Continued		
	Ring elements			Ring elements	
PI(t4)		5	PI(o, 6F)	H(CH ₂) ₀	
PI(c3)		10		o \in {1; 2; 3; 4; 5; 6}	
PI(c4)		15	PI(3xo)	$H(CH_2)_o$ $(CH_2)_oH$	
		20		H(CH ₂) _o	
PI(c5)		25 30	G(o)	$o \in \{1; 2; 3; 4; 5; 6\}$ $H(CH_2)_o \qquad F$	
PI(e5)		35		o ∈ {1; 2; 3; 4; 5; 6}	
DV. O		40	G(i3)	F	
PI(c6)		45	G(t4)	F	
PI(e6)		50	G(c3)		
		55		F	
PI(o, 5F)	H(CH ₂) _o	60	G(c4)	F	
	F o $\in \{1; 2; 3; 4; 5; 6\}$	65			



H—C≡C—

 C_nH_{2n+1} —C=C

N = C - C = C -

-A-

-nA-

-NA-

A	—C=C—	A	_C=C_
V	СН—СН—	V	$-\!\!\operatorname{CH}\!\!=\!\!\operatorname{CH}\!\!-\!\!$
Z	COO	Z	COO
ZI	_O_CO_	ZI	_O_CO_
K	—co—	K	—CO—
W	—CF—CF—	W	—CF—CF—

-A

-An

-AN

Use together with others

—С≡С—Н

 $-C = C - C_n H_{2n+1}$

_C=C_C=N

in which n and m each denote integers, and the three dots "." are placeholders for other abbreviations from this table.

The following table shows illustrative structures together with their respective abbreviations. These are shown in order

to illustrate the meaning of the rules for the abbreviations. They furthermore represent compounds which are preferably used.

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TABLE D

Illustrative structures

The illustrative structures show compounds which are particularly preferably employed.

Examples of compounds of component A

$$C_nH_{2n+1}$$
 $C \equiv C$ $C = C$ C_mH_{2m+1}

 $PTP(0, 5F)TP-n-m, o \ni \{1; 2; 3; 4; 5; 6\}, preferably o = 1 or 2$

$$\mathsf{C}_{n}\mathsf{H}_{2n+1} - \mathsf{C} = \mathsf{C} - \mathsf{C}_{m}\mathsf{H}_{2m+1}$$

 $PTP(o,\,6F)TP\text{-n-m},\,o\ni\{1;\,2;\,3;\,4;\,5;\,6\},\,preferably\,o=1\,\,or\,\,2$

$$C_nH_{2n+1}$$
 $C \equiv C$
 $C = C$
 C_mH_{2m+1}

 $PTP(3xo)TP-n-m, o \ni \{1; 2; 3; 4; 5; 6\}, preferably o = 1$

$$C_nH_{2n+1}$$
 $C \equiv C$
 $C = C$
 C_mH_{2m+1}
 $C = C$
 C_mH_{2m+1}

 $PTP(4xo)TP-n-m, o \ni \{1; 2; 3; 4; 5; 6\}, preferably o = 1$

$$C_nH_{2n+1}$$
 $C \equiv C$ C_mH_{2m+1}

 $PTP(o)TP\text{-n-m}, o \ni \{1; 2; 3; 4; 5; 6\}$

$$C_nH_{2n+1}$$
 $C \equiv C$ $C = C$ $C = C$

 $PTP(o)TP\text{-n-Om, o}\ni\{1;\,2;\,3;\,4;\,5;\,6\}$

Illustrative structures

The illustrative structures show compounds which are particularly preferably employed.

PTP(c5)TP-n-m

Illustrative structures

The illustrative structures show compounds which are particularly preferably employed.

$$C_nH_{2n+1}$$
 $C \equiv C$ $C \equiv C$ $C = C$ C_mH_{2m+1} $C \equiv C$

$$\mathbf{C}_{n}\mathbf{H}_{2n+1} - \mathbf{C} = \mathbf{C} - \mathbf{C}_{m}\mathbf{H}_{2m+1}$$

$$\mathbf{PTP}(\mathbf{e5})\mathbf{TP-n-m}$$

$$\mathbf{C}_{n}\mathbf{H}_{2n+1} - \mathbf{C} = \mathbf{C} - \mathbf{C}_{m}\mathbf{H}_{2m+1}$$

$$\mathbf{PTP(e5)}\mathbf{TP}\text{-n-Om}$$

$$C_nH_{2n+1}$$
 $C \equiv C$ C_mH_{2m+1} $C \equiv C$

$$C_nH_{2n+1}$$
 $C \equiv C$ $C \equiv C$ C_mH_{2m+1} $C \equiv C$

$$C_nH_{2n+1}$$
 $C \equiv C$ $C \equiv C$ $C = C$ $C = C$ $C = C$

Illustrative structures

The illustrative structures show compounds which are particularly preferably employed.

$$C_nH_{2n+1}$$
 $C \equiv C$ $C = C$ $C = C$ $C = C$ $C = C$

$$PTP(o)TP\text{-n-X}, X = F, Cl, o \ni \{1; 2; 3; 4; 5; 6\}$$

$$C_nH_{2n+1}$$
 $C \equiv C$ $C \in C$ $C \in C$ $C \in C$

PTP(o)TU-n-X,
$$X = F$$
, Cl, o $\ni \{1; 2; 3; 4; 5; 6\}$

$$C_nH_{2n+1} - C = C - C - C - NCS$$

$$PTP(o)TP\text{-}n\text{-}S\ o \ni \{1;\, 2;\, 3;\, 4;\, 5;\, 6\}$$

$$\mathsf{C}_n\mathsf{H}_{2n+1} - \mathsf{C} = \mathsf{C} - \mathsf{C}_m\mathsf{H}_{2m+1}$$

 $PTGI(o)TP-n-m, o \ni \{1; 2; 3; 4; 5; 6\}$

$$\mathsf{C}_n\mathsf{H}_{2n+1} - \mathsf{C} = \mathsf{C} - \mathsf{C}_m\mathsf{H}_{2m+1}$$

PTGI(o)TP-n-Om, o ∋ {1; 2; 3; 4; 5; 6}

$$\mathbf{C}_{n}\mathbf{H}_{2n+1} - \mathbf{C} = \mathbf{C} - \mathbf{C}_{m}\mathbf{H}_{2m+1}$$

PTGI(i3)TP-n-m

$$C_nH_{2n+1}$$
 $C \equiv C$
 $C = C$
 C_mH_{2n+1}
 $C = C$

$$\mathbf{C}_{n}\mathbf{H}_{2n+1} - \mathbf{C} = \mathbf{C} - \mathbf{C}_{m}\mathbf{H}_{2m+1}$$

PTGI(c3)TP-n-m

Illustrative structures

The illustrative structures show compounds which are particularly preferably employed.

$$C_nH_{2n+1}$$
 $C \equiv C$
 $C = C$
 $C = C$
 C_mH_{2m+1}
 $C = C$

$$\mathbf{C}_{n}\mathbf{H}_{2n+1} - \mathbf{C} = \mathbf{C} - \mathbf{C}_{m}\mathbf{H}_{2m+1}$$

$$\mathbf{PTGI}(\mathbf{c4})\mathbf{TP-n-m}$$

$$C_nH_{2n+1}$$
 $C \equiv C$
 $C = C$

$$C_nH_{2n+1} - C = C - C_mH_{2m+1}$$

$$PTGI(c5)TP-n-m$$

$$C_nH_{2n+1}$$
 $C \equiv C$
 C_mH_{2n+1}
 $C \equiv C$
 C_mH_{2n+1}

The illustrative structures show compounds which are particularly preferably employed.

F

$$C_nH_{2n+1}$$
 $C \equiv C$
 $C \equiv C$
 C_nH_{2n+1}
 $C \equiv C$
 $C \equiv C$
 $C \equiv C$
 C_nH_{2n+1}
 $C \equiv C$
 $C \equiv C$
 $C \equiv C$
 C_nH_{2n+1}
 $C \equiv C$
 $C \equiv C$

 $PTPP(o)TP\text{-n-m},\,o\ni\{1;\,2;\,3;\,4;\,5;\,6\}$

Illustrative structures

The illustrative structures show compounds which are particularly preferably employed.

$$C_nH_{2n+1} \longrightarrow C \Longrightarrow C \longrightarrow C_mH_{2m+1}$$

 $PTPI(o)P(o)TP-n-m, o \ni \{1; 2; 3; 4; 5; 6\}$

$$\mathsf{C}_n\mathsf{H}_{2n+1} - \mathsf{C} = \mathsf{C} - \mathsf{C}_m\mathsf{H}_{2m+1}$$

 $PTPTP(o)P\text{-n-m},\,o\ni\{1;\,2;\,3;\,4;\,5;\,6\}$

$$C_{n}H_{2n+1} \longrightarrow C \Longrightarrow C \longrightarrow C \Longrightarrow C \longrightarrow C_{m}H_{2m+1}$$

$$\begin{split} & & \text{PTPI}(o)\text{TP}(o)\text{P-n-m, o} \ni \{1; 2; 3; 4; 5; 6\} \\ & \text{(n} \ni \{1; 2; 3; 4; 5; 6; 7\} \text{ and m} \ni \{1; 2; 3; 4; 5; 6; 7\}.) \end{split}$$

Examples of compounds of component B

$$C_nH_{2n+1}$$
 CF_2-O F

PGUQU-n-F

$$C_nH_{2n+1}$$
 CF_2-O F CF_2-O F

PUQGUQU-n-F

 $PUQGUQGU-n-F = PU[QGU]_2-n-F$

$$\begin{split} &PUQGU(m)QGU\text{-}n\text{-}F = PU[QGU]_2^{(m)}\text{-}n\text{-}F \\ &(n \ni \{1;\, 2;\, 3;\, 4;\, 5;\, 6;\, 7\} \text{ and } m \ni \{1;\, 2;\, 3;\, 4\}.) \end{split}$$

Illustrative structures

The illustrative structures show compounds which are particularly preferably employed.

Examples of compounds of component C

$$C_nH_{2n+1}$$
 C_mH_{2m+1}
 C_mH_{2m+1}

$$C_nH_{2n+1}$$
 O C_mH_{2m+1}

 $\begin{array}{c} {\rm PfX\text{-}n\text{-}Om} \\ (n\ni\{1;\,2;\,3;\,4;\,5;\,6;\,7\} \ {\rm and} \ m\ni\{1;\,2;\,3;\,4;\,5;\,6;\,7\}) \end{array}$

Examples of compounds of component E Compounds having three 6-membered rings

$$\mathbf{C}_{n}\mathbf{H}_{2n+1} - \mathbf{C}_{m}\mathbf{H}_{2m+1}$$

$$\mathbf{C}_{n}\mathbf{H}_{2n+1} - \mathbf{C}_{n}\mathbf{H}_{2n+1} - \mathbf{C}_{n}\mathbf{H}_{2n+1} - \mathbf{C}_{n}\mathbf{H}_{2n+1}$$

Illustrative structures

The illustrative structures show compounds which are particularly preferably employed.

(n \ni {1; 2; 3; 4; 5; 6; 7}, m \ni {1; 2; 3; 4; 5; 6; 7}, and k \ni {0; 1; 2; 3; 4}, preferably 0 or 2, and l \ni {0; 1; 2; 3}.)

Compounds having four 6-membered rings

 $\begin{array}{c} PGIYP\text{-}n\text{-}m \\ (n\ni\{1;2;3;4;5;6;7\} \text{ and } m\ni\{1;2;3;4;5;6;7\}) \end{array}$

TABLE D-continued

Illustrative structures
The illustrative structures show compounds which are particularly preferably employed.

Illustrative structures of polar compounds employed:

$$C_nH_{2n+1}$$

PGU-n-F

$$C_nH_{2n+1}$$

PPGU-n-F

$$C_nH_{2n+1}$$
 CF_2 CF_2 CF_2 F

PUQU-n-F (n \ni {1; 2; 3; 4; 5; 6; 7}.)

Illustrative structures of further neutral compounds employed:

$$C_nH_{2n+1}$$
 C_mH_{2m+1} C_mH_{2m+1}

CCZPC-n-m

$$\Gamma_{n}$$
 Γ_{n} Γ_{n

CPGP-n-m

$$C_nH_{2n+1}$$
 $CH_2)_k$ CH_2

CPGP-n-kV

Illustrative structures
The illustrative structures show compounds which are particularly preferably employed.

$$C_nH_{2n+1}$$
 C_nH_{2n+1} C_nH_{2n+1} C_nH_{2n+1}

CPGP-n-kVl

 $(n \ni \{1; 2; 3; 4; 5; 6; 7\}, m \ni \{1; 2; 3; 4; 5; 6; 7\}, and k \ni \{0; 1; 2; 3; 4\}, preferably 0 or 2, and 1 \ni \{0; 1; 2; 3\}.)$

Illustrative structures of further polar compounds employed:

$$C_nH_{2n+1}$$

CGU-n-F

$$C_nH_{2n+1}$$

CCPU-n-F

$$C_nH_{2n+1}$$
 F

CCGU-n-F

$$C_nH_{2n+1}$$

CPGU-n-F

$$C_nH_{2n+1}$$
 OCF₃

CPGU-n-OT

Illustrative structures

The illustrative structures show compounds which are particularly preferably employed.

$$C_nH_{2n+1}$$
 OCF₃
 CCP -n-OT

 C_nH_{2n+1} OCF₃
 $CCEP$ -n-OT

 C_nH_{2n+1} F

 CPG -n-F

 $(n \ni \{1; 2; 3; 4; 5; 6; 7\}.)$

The following table, Table E, shows illustrative compounds which can be used as stabiliser in the mesogenic media in accordance with the present invention. The total concentration of these and similar compounds in the media is $_{\rm 30}$ preferably 5% or less.

TABLE E

$$C_nH_{2n+1}$$
 OH

55

TABLE E-continued

$$C_nH_{2n+1}$$
 H OH $C_nH_{2n+1}OH$ OH OH

15

TABLE E-continued

$$\begin{array}{c} O - C_8 H_{17} \\ \\ O \\ \\ O \\ \end{array}$$

$$\begin{array}{c|c} & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ &$$

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In a preferred embodiment of the present invention, the mesogenic media comprise one or more compounds selected from the group of the compounds from Table E.

The following table, Table F, shows illustrative compounds which can preferably be used as chiral dopants in the mesogenic media in accordance with the present invention.

TABLE F

$$C_{2}H_{5} \stackrel{*}{\overset{\circ}{\subset}} H - CH_{2}O \longrightarrow CN$$

$$C_{2}H_{5} \stackrel{*}{\overset{\circ}{\subset}} H - CH_{2} \longrightarrow CN$$

$$C_{2}H_{5} \stackrel{*}{\overset{\circ}{\subset}} H - CH_{2} \longrightarrow CN$$

$$C_{6}H_{13} \stackrel{*}{\overset{\circ}{\subset}} H - O \longrightarrow C_{3}H_{11}$$

$$C_{3}H_{7} \longrightarrow H \longrightarrow CH_{2} \stackrel{*}{\overset{\circ}{\subset}} H - C_{2}H_{5}$$

$$C_{4}H_{17}O \longrightarrow CM 45$$

$$C_{5}H_{17}O \longrightarrow CM 47$$

$$H_3C$$
 H_3C
 CH_3
 CH_3
 CH_3

$$\begin{array}{c} H_3C\\ H_3C\\ \end{array}$$

$$C_6H_{13}O - C_6H_{13}O - C_6H_{13}O - C_{13}O - C_{13$$

In a preferred embodiment of the present invention, the mesogenic media comprise one or more compounds selected from the group of the compounds from Table F.

The mesogenic media in accordance with the present application preferably comprise two or more, preferably four or more, compounds selected from the group consisting of the compounds from the above tables.

The liquid-crystal media in accordance with the present invention preferably comprise

seven or more, preferably eight or more, compounds, preferably compounds having three or more, preferably four or more, different formulae, selected from the group of the compounds from Table D.

EXAMPLES

The following examples illustrate the present invention without limiting it in any way. However, it becomes clear to the person skilled in the art from the physical properties what properties can be achieved and in what ranges they can be modified. In particular, the combination of the various properties which can preferably be achieved is thus well defined for the person skilled in the art.

The acetylenes employed, if not commercially available, are synthesised in accordance with standard laboratory procedures.

Substance Examples A

Illustrative Compounds of the Formula IA for Component A

Synthesis Example 1a

1,4-bis(4-n-Butylphenylethynyl)-2,5-dimethylbenzene

$$Br$$
 Br
 C_4H_9
 E_1
 E_2
 E_3
 E_4
 E_4

9 g (54 mmol) of 4-n-butylphenylacetylene are initially introduced in 200 ml of tetrahydrofuran, the mixture is cooled to -75° C., and 54 ml (1.0 M in hexane, 54 mmol) of lithium hexamethyldisilazide are added. After 1 h, 54 ml (1.0 M in hexane, 54 mmol) of B-methoxy-9-borabicyclo[3.3.1] nonane are added, and the mixture is stirred for a further hour.

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This solution is added dropwise to a solution of 1.5 g (2 mmol) of bis(tricyclohexylphosphine)palladium(II) chloride and 5.3 g (20 mmol) of 1,4-dibromo-2,5-dimethylbenzene in 300 ml of tetrahydrofuran, and the mixture is refluxed for 16 h

The mixture is worked up by addition of MTB and water, the phases are separated, and the aqueous phase is extracted once with MTB. The combined organic phases are washed with saturated sodium chloride soln., dried over sodium sulfate, filtered and evaporated in a rotary evaporator. The residue is purified by column chromatography (SiO₂, heptane/chlorobutane=1:1); the further purification is carried out by recrystallisation from heptane or heptane/chlorobutane=1:1.

MS (EI): m/e (%)=418 (100, M⁺), 375 (26, [M-propyl]⁺), 166 (9, [M-2x propyl]²⁺).

Phase sequence: C 125° C. N 142.8° C. I. Data extrapolated from 10% solution in ZLI-4792: $\Delta \epsilon = +2.6$; $\Delta n = 0.414$ and $\gamma_1 = 2080$ mPa·s.

Synthesis Example 2a

1,4-bis(4-n-Butylphenylethynyl)-2,6-dimethylbenzene

1,4-bis(4-n-Butylphenylethynyl)-2,6-dimethylbenzene is prepared from 4-n-butylphenylacetylene and 1-iodo-4-bromo-2,6-dimethylbenzene analogously to the synthesis described under 1A.

MS (EI): m/e (%)=418 (100, M⁺), 375 (27, [M-propyl]⁺), 166 (10, [M-2x propyl]²⁺).

Phase sequence: C 115° C. N 145.8° C. I. Data extrapolated from 10% solution in ZLI-4792: $\Delta \epsilon$ =+2.4; Δ n=0.409 and γ ₁=1940 mPa·s.

Synthesis Example 3a

1-(4-n-Propylphenylethynyl)-2-fluoro-4-(4-n-hexylphenylethynyl)-6-ethylbenzene

$$H_{13}C_6$$
 C_3H_7

3a.1 Synthesis of rac-2-fluoro-6-(1-hydroxyethyl)-1,4-dibromobenzene

 $120\,g$ (0.36 mol) of 6-fluoro-1,2,4-tribromobenzene in 120 ml of tetrahydrofuran are added dropwise to 290 ml (0.38 $_{35}$ mol, 1.3 M solution in tetrahydrofuran) of isopropylmagnesium chloride/lithium chloride complex cooled to -70° C., and the mixture is stirred at -50° C. for 2 h. The mixture is subsequently cooled to -70° C., before 17.6 g (0.4 mol) of acetaldehyde in 40 ml of tetrahydrofuran are added dropwise. $_{40}$ The batch is slowly warmed to -20° C. and then quenched by the addition of saturated ammonium chloride solution.

The phases are separated, and the aqueous phase is extracted with MTB. The combined organic phases are washed with saturated sodium chloride solution, dried over 45 sodium sulfate and evaporated in a rotary evaporator. The residue obtained in this way is purified a number of times by column chromatography (SiO₂, heptane→(gradient) heptane/MTB=4:1), with the polarity of the eluent being increased successively by the addition of MTB.

3a.2 Synthesis of 2-fluoro-6-ethyl-1,4-dibromobenzene

$$B_{r}$$
 B_{r}
 $E_{t_{3}Si}$
 B_{r}
 B_{r}
 B_{r}
 B_{r}
 B_{r}
 B_{r}

 $11~{\rm g}$ (36.9 mmol) of the alcohol obtained in the preceding reaction are initially introduced in 25 ml of dichloromethane,

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8.7 g (75.2 mmol) of triethylsilane are added, and the mixture is stirred at room temperature for 3 h. 10.4 g (72.9 mmol) of boron trifluoride diethyl etherate are subsequently added dropwise, and the mixture is stirred at a temperature just below the boiling point for 12 days. The batch, which is then concentrated virtually completely, is cooled, MTB and 2 M sodium hydroxide solution are added, and the mixture is stirred vigorously for 3 h. The phases are separated, and the aqueous phase is extracted twice with MTB. The combined organic phases are washed with saturated sodium chloride solution, dried over sodium sulfate and evaporated in a rotary evaporator. The residue obtained in this way is purified by column chromatography (SiO₂, heptane).

3a.3 Synthesis of 1-bromo-2-fluoro-4-(4-n-hexylphenylethynyl)-6-ethylbenzene

$$\begin{array}{c} F \\ Br \\ \hline \\ Br \\ \hline \\ Cu(I)/Pd(0) \\ \hline \\ C_6H_{13} \\ \hline \\ C_6H_{13} \\ \hline \\ \end{array}$$

 $3.4\,\mathrm{g}\,(12\,\mathrm{mmol})$ of the dibromide obtained in the preceding reaction, $2.6\,\mathrm{g}\,(13.9\,\mathrm{mmol})$ of 4-n-hexylphenylacetylene, $0.5\,\mathrm{g}\,(0.7\,\mathrm{mmol})$ of bis(triphenylphosphine)palladium(II) chloride and 70 mg (0.35 mmol) of copper(I) iodide are initially introduced in 70 ml of diisopropylamine, and the mixture is stirred at 50° C. for 1 h and under reflux for 1 h. The batch is cooled, and the precipitate formed is filtered off. The filtrate is evaporated in a rotary evaporator, and the residue obtained in this way is purified by column chromatography (SiO $_2$, heptane).

3a.4 Synthesis of 1-(4-n-propylphenylethynyl)-2-fluoro-4-(4-n-hexylphenylethynyl)-6-ethylbenzene

2 g (5.2 mmol) of the bromide obtained in the preceding reaction, 1.2 g (8.3 mmol) of 4-n-propylphenylacetylene, 0.2 g (0.28 mmol) of bis(triphenylphosphine)palladium(II) chloride and 30 mg (0.16 mmol) of copper(I) iodide are initially

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35

40

50

1204a.2 Synthesis of 1-(4-n-propylphenylethynyl)-2-fluoro-4-(4-n-hexylphenylethynyl)-6-ethylbenzene

introduced in 30 ml of diisopropylamine, and the mixture is heated under reflux. After 1 h and after 12 h, a further 1.2 g of 4-n-propylphenylacetylene in a little diisopropylamine are added in each case, before the mixture is cooled after a total of 60 h, and the precipitate formed is filtered off. The filtrate is evaporated in a rotary evaporator, and the residue obtained in this way is purified by column chromatography (SiO₂, heptane); the further purification is carried out by recrystallisation from heptane.

Phase sequence: C 33° C. N 90° C. I. Data extrapolated from 10% solution in ZLI-4792: $\Delta\epsilon{=}{+}1.8; \Delta n{=}0.389 \text{ and } \gamma_1{=}1530 \text{ mPa}{\cdot}\text{s}.$

Synthesis Example 4a

1-(4-n-Propylphenylethynyl)-2-fluoro-4-(4-n-hexylphenylethynyl)-6-ethylbenzene

$$H_{13}C_6$$
 C_3H_7

4a.1 Synthesis of 2-(4-n-propylphenylethynyl)-4fluoro-5-bromobenzene

$$I \longrightarrow Br \longrightarrow Cu(I)/Pd(0)$$

$$H_7C_3 \longrightarrow Br$$

 $20\,\mathrm{g}$ (63.5 mmol) of 5-bromo-4-fluoro-2-iodotoluene, 10.5 g (73 mmol) of 4-n-propylphenylacetylene, 2.5 g (3.6 mmol) of bis(triphenylphosphine)palladium-(II) chloride and 300 $_{60}$ mg (1.8 mmol) of copper(I) iodide are initially introduced in 250 ml of diisopropylamine, the mixture is heated under reflux for 3 h and cooled, and the precipitate formed is filtered off. The filtrate is evaporated in a rotary evaporator, and the residue obtained in this way is purified by column chromatography (SiO $_{2}$, heptane).

$$H_7C_3 \longrightarrow F$$

$$E_7C_3 \longrightarrow C_6H_{13}$$

$$C_0(I)/Pd(0)$$

$$C_6H_{13}$$

 $9.6\,\mathrm{g}$ (28.9 mmol) of the bromide obtained in the preceding reaction, 7 g (37.6 mmol) of 4-n-hexylphenylacetylene, 1.7 g (2.4 mmol) of bis(triphenylphosphine)palladium(II) chloride and 200 mg (1.2 mmol) of copper(I) iodide are initially introduced in 250 ml of diisopropylamine, the mixture is heated under reflux for 2 h and cooled, and the precipitate formed is filtered off. The filtrate is evaporated in a rotary evaporator, and the residue obtained in this way is purified by column chromatography (SiO_2, heptane); the further purification is carried out by recrystallisation from heptane.

MS (EI): m/e (%)=436 (100, M⁺), 407 (8, [M-ethyl]⁺), 365 (39, [M-pentyl]⁺), 336 (12).

Phase sequence: C 50° C. N 148.7° C. I.

Data extrapolated from 10% solution in ZLI-4792: $\Delta \epsilon = +2.3; \Delta n = 0.427 \text{ and } \gamma_1 = 1490 \text{ mPa·s}.$

Synthesis Example 5a

Synthesis of 1,4-bis-(4-n-butylphenylethynyl)-2,3,5, 6-tetramethylbenzene

$$I \longrightarrow Cu(I)/Pd(0)$$

$$H_9C_4 \longrightarrow C_4H$$

 $8.1\,\mathrm{g}$ (20 mmol) of 1,4-diiodo-2,3,5,6-tetramethylbenzene, 7.5 g (47.3 mmol) of 4-n-butylphenylacetylene, 2 g (2.9 mmol) of bis(triphenylphosphine)palladium(II) chloride and 550 mg (2.9 mmol) of copper(I) iodide are initially introduced in 100 ml of triethylamine, and the mixture is heated under reflux for 3 h. The batch is then cooled, water and MTB are added, and the phases are separated. The organic phase is washed with saturated ammonium chloride solution and with sodium chloride solution and evaporated in a rotary evaporator, and the residue obtained in this way is purified by column chromatography (SiO₂, heptane); the further purification is carried out by recrystallisation from isopropanol.

MS (EI): m/e (%)=446 (100, M⁺), 421 (15, [M-propyl]⁺), 180 (8, [M-2x propyl]²⁺).

Phase sequence: C 67° C. N 225.1° C. I. Data extrapolated from 10% solution in ZLI-4792: $\Delta \epsilon = +2.3$; $\Delta n = 0.423$ and $\gamma_1 = 1360$ mPa·s.

121 Synthesis Example 6a

1,4-bis(4-n-Butylphenylethynyl)-2,3,5-trimethylbenzene

$$C_4H_9 \longrightarrow C_4H_9 \longrightarrow C$$

6a.1: Synthesis of 4-bromo-2,3,6-trimethylphenol

6a.3: Synthesis of 1,4-bis-(4-n-butylphenylethynyl)-2,3,5-trimethylbenzene

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$$\begin{array}{c|c} & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & &$$

10 g (71 mmol) of 2,3,6-trimethylphenol are initially introduced in a solvent mixture consisting of 250 ml of dichloromethane and 190 ml of methanol, and 35.4 g (72 mmol) of tetrabutylammonium tribromide in 170 ml of dichloromethane and 130 ml of methanol are added. After 5 h at room temperature, the batch is evaporated in a rotary evaporator, and the residue remaining is taken up in MTB. The organic solution is washed three times with water, dried over sodium sulfate and evaporated in a rotary evaporator. The crystalline residue obtained is employed in the next step without further work-up.

6a.2: Synthesis of trifluoromethylsulfonyl-4-bromo-2,3,6-trimethylphenol

$$\begin{array}{c|c} & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\$$

15.5~g~(69~mmol) of the phenol obtained in the preceding reaction are initially introduced in 230~ml of dichloromethane, 14.7~ml~(106~mmol) of triethylamine and 200~mg~60~(1.6~mmol) of 4-(dimethylamino)pyridine are added, and the mixture is cooled in an ice bath. 12.2~ml~(72.3~mmol) of trifluoromethanesulfonic anhydride are then added dropwise, and the mixture is stirred overnight and warmed to room temperature in the process, before the entire batch is purified directly by column chromatography (SiO $_2$, dichloromethane).

$$\begin{array}{c} \text{Br} \\ \\ \text{OSO}_2\text{CF}_3 \end{array} \xrightarrow[\text{LiHMDS, 9-MeO-9-BBN}]{} \\ \text{Pd}(0) \\ \end{array}$$

$$H_9C_4$$
 C_4H_9

9 g (54 mmol) of 4-n-butylphenylacetylene are initially introduced in 100 ml of tetrahydrofuran, the mixture is cooled to -78° C., and 54 ml (1.0 M in hexane, 54 mmol) of lithium hexamethyldisilazide are added. After 1 h, 54 ml (1.0 M in 45 hexane, 54 mmol) of B-methoxy-9-borabicyclo[3.3.1] nonane are added, and the mixture is stirred for a further 2 h. The batch is then warmed to -20° C., 730 mg (0.8 mmol) of tris(dibenzylideneacetone)dipalladium(0), 1.3 g (3.2 mmol) of 2-dicyclohexylphosphino-2',6'-dimethoxybiphenyl and 6.9 g (20 mmol) of the triflate obtained in the preceding reaction in 150 ml of tetrahydrofuran are added, and the mixture is heated under reflux for 16 h. After cooling, the mixture is worked up by addition of MTB and water, the phases are separated, and the aqueous phase is extracted once with MTB. The combined organic phases are washed with saturated sodium chloride solution, dried over sodium sulfate, filtered and evaporated in a rotary evaporator. The residue is purified by column chromatography (SiO₂, heptane); the final purification is carried out by recrystallisation from heptane.

MS (EI): m/e (%)=432 (100, M⁺), 389 (20, [M-propyl]⁺), 173 (8, [M-2x propyl]²⁺).

Phase sequence: C 136° C. N 184.2° C. I. Data extrapolated from 10% solution in ZLI-4792: $\Delta \epsilon$ =+2.4; Δn =0.412 and γ_1 =1650 mPa·s.

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Substance Examples B

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Substance Example 6b

Illustrative Compounds of the Formula IB for Component A

Substance Example 1b

$$C_4H_9$$
 $C \equiv C$ C_4H_9

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 C_4H_9 $C \equiv C$ C_4H_9

Phase sequence: C 29° C. N 119.2° C. I; Δ n=0.402; Δ ϵ =1.7.

Substance Example 7b

Phase sequence: C 87° C. N 178.5° C. I; Δ n=0.436; Δ ϵ =2.8.

Substance Example 2b

 C_3H_7 $C \equiv C$ C_6H_1

$$C_{6}H_{13}$$
 $C \equiv C$ $C_{3}H_{7}$

Phase sequence: T_g –54° C. C 14° C. N 119.2° C. I; $\Delta n=0.393$: $\Delta \epsilon=1.8$.

Phase sequence: C78° C. N 172.3° C. I; Δ n=0.437; Δ ϵ =2.6. $_{30}$

Substance Example 8b

Substance Example 3b

$$C_5H_{11}$$
— C = C — C 3 H_7

 C_6H_{13} $C\equiv C$ C_3H_7

Phase sequence: C 107° C. N 211.0° C. I; $\Delta n{=}0.464;$ $\Delta \varepsilon{=}3.1.$

Phase sequence: C 60° C. N 121.8° C. I; Δ n=0.394; Δ ϵ =1.7.

Substance Example 9b

Substance Example 4b

$$C_2H_5$$
 $C \equiv C$ C_2H_5

$$C_5H_{11}$$
— C = C — C 3 H_7

Phase sequence: C 87° C. N 130.7° C. I; $\Delta n=0.451$; $\Delta \epsilon=2.1$. 55

Phase sequence: C81° C. N 160.7° C. I; Δ n=0.432; Δ ϵ =3.2.

Substance Example 5b

Substance Example 10b

$$C_3H_7$$
 $C \equiv C$ C_3H_7

60
$$C_4H_9 - C \equiv C - C_4H_9$$

Phase sequence: C 57° C. N 151.7° C. I; $\Delta n=0.445$; $\Delta \epsilon=2.0$.

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Synthesis Example 10:1,4-Bis(2-(4-butylphenyl) ethynyl)-2-cyclopropylbenzene

10b.1) 1,4-Dichloro-2-cyclopropylbenzene 2

 $20~g~(73~mmol)~of~1,4-dichloro-2-iodobenzene, 9.4~g~(110~mmol)~of~cyclopropylboronic~acid, <math display="inline">32~g~(147~mmol)~of~potassium~phosphate, <math display="inline">421~mg~(0.7~mmol)~of~bis(dibenzylideneacetone)~palladium(0)~(Pd(dba)_2)~and~1096~mg~(1.5~mmol)~of~25~1,2,3,4,5-pentaphenyl-1-(di-tert-butylphosphine)~ferrocene~(CTC-Q-PHOS)~are~dissolved~in~600~ml~of~toluene~and~heated~at~100°~C.~overnight.~100~ml~of~water~are~added~to~the~cooled~solution,~and~the~mixture~is~extracted~twice~with~toluene~(100~ml).~The~combined~organic~phases~are~washed~with~water,~dried~over~sodium~sulfate~and~evaporated~in~vacuo.~The~residue~is~purified~by~column~chromatography,~giving~the~title~compound~as~a~colourless~solid.$

10b.2) 1,4-Bis(2-(4-butylphenyl)ethynyl)-2-cyclopropylbenzene (1)

$$C_1$$
 C_1
 C_4
 C_4
 C_4
 C_4
 C_4
 C_4

5 g (26 mmol) of 1,4-dichloro-2-cyclopropylbenzene, 9.4 g (58 mmol) of 1-n-butyl-4-ethynylbenzene, 19 g (58 mmol) of caesium carbonate, 69 mg (0.3 mmol) of bis(acetonitrile) palladium(II) chloride and 382 mg (0.8 mmol) of 2-dicyclo-55 hexylphosphino-2',4',6'-triisopropylbiphenyl are dissolved in 80 ml of dioxane under nitrogen, and the mixture is heated at 100° C. overnight. 100 ml of water are added to the cooled solution, and the mixture is extracted twice with methyl t-butyl ether (100 ml). The combined organic phases are washed with water, dried over sodium sulfate and evaporated in vacuo. The residue is purified by column chromatography and recrystallised from ethanol, giving the title compound I as a solid.

 $^{1}\mbox{H-NMR}$ (250 MHz, CDCl3): 7.4-7.08 (11H, m); 2.5 (4H, 65 m); 1.6-1.3 (9H, m); 0.96 (6H, m); 0.6-0.4 (4H, m).

Phase sequence: C 72° C. N 84.5° C. I; Δ n=0.378; Δ ϵ =1.5.

$$C_3H_7$$
— $C\equiv C$ — $C=C$ — C_6H_{13}

Phase sequence: T $_g$ –43° C. C 46° C. N 86.0° C. I; $\Delta n{=}0.379; \,\Delta \epsilon{=}1.1.$

Substance Example 12b

20
 $_{\text{C}_{4}\text{H}_{9}}$ $_{\text{C}}$ $_{\text{C}}$ $_{\text{C}}$ $_{\text{C}}$ $_{\text{C}}$

Synthesis Example 12b

1,4-Bis(2-(4-butylphenyl)ethynyl)-2-cyclobutylbenzene (3)

$$H_9C_4$$
 C_4H_9

12b.1) 1-(2,5-Dibromophenyl)cyclobutanol (4)

$$\operatorname{Br}$$
 Br
 HO
 Br
 Br
 HO
 Br
 Br
 A

21.09 g (67 mmol) of 1,2,4-tribromobenzene are initially introduced in 100 ml of THF under nitrogen, cooled to -45° C., and a solution (1.3 M) of 51.54 ml (67 mmol) of isopropylmagnesium chloride/lithium chloride complex in THF is added dropwise. After 1 hour, the batch is warmed to -10° C., and 5 ml (66.34 ml) of cyclobutanone are added dropwise at this temperature. The batch is allowed to thaw, and sat. NH₄Cl solution is added, the mixture is extracted with methyl t-butyl ether, the organic phase is dried over sodium sulfate and filtered, the solvent is removed in vacuo, and the residue is

filtered through silica gel with dichloromethane, giving 4, which is employed in the next step without further purification.

12b.2) 1,4-Dibromo-2-cyclobutylbenzene (5)

14.5 g (47.39 mmol) of 4 are dissolved in 50 ml of THF under nitrogen, and 35.72 ml (284.4 mmol) of boron trifluoride/diethyl ether complex are added dropwise at room temperature, and 12.54 g (189.6 mmol) of sodium cyanoborohydride are added dropwise in portions. The batch is heated under reflux overnight. The batch is allowed to cool to room temperature, sat. NaHCO₃ solution is added, the mixture is extracted with methyl t-butyl ether, the organic phase is dried over sodium sulfate and filtered, the solvent is removed in vacuo, and the residue is filtered through silica gel with 1-chlorobutane, giving 5 as a yellow liquid.

12b.3) 1,4-Bis(2-(4-butylphenyl)ethynyl)-2-cyclobutylbenzene (3)

7.8 g (47.0 mmol) of 1-butyl-4-ethynylbenzene are ini- 50 tially introduced in 100 ml of THF under nitrogen, cooled to -78° C., and 63.32 ml (63.20 mmol) of a 1 M solution of lithium bis(trimethylsilyl)amide in hexane are added dropwise. After 1 hour, 63.22 ml (63.20 mmol) of a 1 M solution of 9-methoxy-9-BBN in hexane are added, and the mixture is 55 left to stir at -78° C. for 2 hours. In a second apparatus, 6.8 g (23.45 mmol) of 5, 0.916 g (1.0 mmol) of tris(dibenzylideneacetone)dipalladium(0) and 1.64 g (4.0 mmol) of 2-dicyclohexylphosphino-2',6'-dimethoxybiphenyl in 100 ml of THF are initially introduced. The first solution is slowly 60 added dropwise, and the batch is heated at 100° C. overnight. 100 ml of water are added to the cooled solution, and the mixture is extracted twice with methyl t-butyl ether (100 ml). The combined organic phases are washed with water, dried over sodium sulfate and evaporated in vacuo. The residue is 65 purified by column chromatography and recrystallised from isopropanol, giving the title compound 3 as a solid.

Phase sequence: T $_g$ –39° C. C 69° C. N 70.1° C. I; Δ n=0.359; Δ e=0.9.

Substance Example 13b

$$C_4H_9$$
 $C \equiv C$ C_4H_5

Phase sequence: T $_g$ –36° C. C 57° C. N(?) (7.0)° C. I; 15 $\Delta n{=}0.334;$ $\Delta \epsilon{=}{-}0.1.$

Substance Example 14b

$$C_4H_9$$
— $C\equiv C$ — $C=C$ — C_4H_9

Phase sequence: T $_g$ –34° C. C 47° C. N 53.1° C. I; $\Delta n{=}0.337; \Delta \epsilon{=}0.0.$

Substance Example 15b

$$C_6H_{13}$$
 $C \equiv C$ $C \equiv C$

Synthesis Example 15b

2-Cyclohexyl-4-(4-hexylphenylethynyl)-1-(4-propylphenylethynyl)benzene 6

15b.1) 4-Chloro-2-cyclohexylbenzene trifluoromethanesulfonate (7)

19 g (90.2 mmol) of 4-chloro-2-cyclohexylbenzene are dissolved in 264 ml of dichloromethane, cooled to -5° C., and 4.64 ml (33.18 mmol) of triethylamine and 223 mg (1.8 mmol) of 4-(dimethylamino)pyridine are added dropwise. The batch is stirred at room temperature overnight and filtered

15

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through silica gel with dichloromethane, giving the product 7, which is employed in the next step without further purification.

15b.2) (4-Chloro-2-cyclohexylphenylethynyl)trimethylsilane (8)

21 g (61.3 mmol) of 7, 25.8 ml (183.8 mmol) of trimethylsilylacetylene, 2.15 g (3 mmol) of bis(triphenylphosphine) palladium(II) chloride and 21.2 ml (153.2 mmol) of triethylamine are dissolved in 60 ml of N,N-dimethylformamide under nitrogen, and the mixture is heated at 100° C. overnight. 100 ml of water are added to the cooled solution, and the mixture is extracted twice with methyl t-butyl ether (100 ml). The combined organic phases are washed with water, dried over sodium sulfate and evaporated in vacuo. The residue is purified by column chromatography, giving the product 8, 40 which is employed in the next step without further purification.

15b.3) 4-Chloro-2-cyclohexyl-1-ethynylbenzene (9)

$$CI$$
 \longrightarrow $Si(CH_3)_3$ \longrightarrow $Si(CH_3)_3$ \longrightarrow $Si(CH_3)_4$ \longrightarrow $Si(CH_3)_4$ \longrightarrow $Si(CH_3)_4$ \longrightarrow $Si(CH_3)_5$ \longrightarrow \longrightarrow $Si(CH_3)_5$ \longrightarrow $Si(CH_3)_5$ \longrightarrow $Si(CH_3)_5$ \longrightarrow $Si(CH_3)$

16.6~g~(57.1~mmol)~of~8~are dissolved in 154 ml of tetrahy- $_{65}$ drofuran, cooled to 0° C., and a 1 M solution of tetra-n-butylammonium fluoride (68.48 mmol) is added dropwise.

The batch is stirred at room temperature overnight, water is added, the mixture is extracted with methyl t-butyl ether, the organic phase is dried over sodium sulfate and filtered, the solvent is removed in vacuo, and the residue is filtered through silica gel with heptane/toluene, giving the product 9, which is employed in the next step without further purification.

15b.4) 4-Chloro-2-cyclohexyl-1-p-tolylethynylbenzene (10)

CI—H

Br

$$C_6H_{13}$$
 C_6H_{13}
 C_6H_{13}

6.6 g (30.17 mmol) of 9, 7.28 g (30.17 mmol) of 1-bromo-4-hexylbenzene, 21.63 g (66.39 mmol) of caesium carbonate, 78 mg (0.3 mmol) of bis(acetonitrile)palladium(II) chloride and 431 mg (0.9 mmol) of 2-dicyclohexylphosphino-2',4',6'-triisopropylbiphenyl are dissolved in 90 ml of dioxane under nitrogen and heated at 100° C. overnight. 100 ml of water are added to the cooled solution, and the mixture is extracted twice with methyl t-butyl ether (100 ml). The combined organic phases are washed with water, dried over sodium sulfate and evaporated in vacuo. The residue is purified by column chromatography.

15b.5) 2-Cyclohexyl-4-(4-hexylphenylethynyl)-1-(4-propylphenylethynyl)benzene (6)

$$C_{6}H_{13}$$

25

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55

-continued
$$H_{13}C_{6}$$

$$C_{3}H_{7}$$

4.5 g (11.87 mmol) of 10, 1.7 g (11.87 mmol) of 1-n-propyl-4-ethynylbenzene, 8.5 g (26.12 mmol) of caesium carbonate, 30 mg (0.1 mmol) of bis(acetonitrile)palladium (II) chloride and 170 mg (0.35 mmol) of 2-dicyclohexylphosphino-2',4',6'-triisopropylbiphenyl are dissolved in 35 ml of dioxane under nitrogen and heated at 100° C. overnight. 100 ml of water are added to the cooled solution, and the mixture is extracted twice with methyl t-butyl ether (100 ml). The combined organic phases are washed with water, dried over sodium sulfate and evaporated in vacuo. The residue is purified by column chromatography, giving the title compound 6 as a solid.

Phase sequence: $T_g -23^{\circ}$ C. I; $\Delta n=0.294$; $\Delta \epsilon=-0.6$.

Substance Example 16b

$$C_4H_9$$
 $C \equiv C$ C_4H_9

The compound is prepared in accordance with Scheme $10_{\ 35}$ and analogously to compounds already described in the present application.

Phase sequence: C 90° C. N 193.9° C. I; Δ n=0.435; Δ ϵ =3.0.

Substance Example 17b

$$C_3H_7$$
 $C \equiv C$ C_6H_{13}

The compound is prepared in accordance with Scheme 10 and analogously to compounds already described in the present application. 50

Phase sequence: C 53° C. N 143.5° C. I; Δ n=0.401; $\Delta\epsilon$ =2.5.

Substance Example 18b

$$C_4H_9$$
 $C \equiv C$ C_4H_9 $C \equiv C$

The compound is prepared in accordance with Scheme 10 65 and analogously to compounds already described in the present application.

Phase sequence: T $_g$ –49 C 32° C. N 126.0° C. I; Δ n=0.373; $\Delta\epsilon$ =1.6.

Substance Example 19b

$$C_4H_9$$
 $C \equiv C$ C_3H_7

Substance Example 20b

$$C_4H_9$$
 $C\equiv C$
 C_4H_9
 $C\equiv C$
 C_4H_9

Phase sequence: T $_g$ –45° C. C 60° C. N 89.1° C. I; Δ n=0.348; Δ e=1.3.

Substance Example 21b

$$C_4H_9$$
 $C \equiv C$
 $C \equiv C$
 C_4H_9

Phase sequence: T $_g$ –32° C. C 66° C. N (44.1)° C. I; $\Delta n{=}0.322; \Delta \epsilon{=}0.6.$

Substance Example 22b

$$C_4H_9$$
 $C \equiv C$
 $C \equiv C$
 C_4H_9

Phase sequence: T_g -37° C. C 52° C. N 78.2° C. I; Δ n=0.339; Δ e=0.9.

Synthesis Example 23b

$$C_4H_9$$
 C_4H_9 C_4H_9

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$$\begin{array}{c|c} & & & & Br \\ & & & C_2H_5 \\ \hline & & & & C_1 \\ \hline \end{array}$$

25.4 g of 2-bromo-1,4-dichlorobenzene, 8.9 ml of iodoethane and 13.3 ml of DMPU are dissolved in 230 ml of THF, and a solution of 16.2 g of lithium tetramethylpiperidite in THF is added dropwise at -70° C. After a further 2 h at -70° C., the reaction mixture is allowed to warm to ambient temperature, and the batch is hydrolysed using water and subjected to extractive work-up.

The crude product is purified by fractional distillation. B.p.: 73° C./0.1 bar. Colourless liquid.

Step 23b.2

12.5 ml of a 5% solution of methyllithium in diethyl ether are added to 2.4 g of anhydrous zinc bromide in 50 ml of THF at 25-40° C. 0.3 g of $PdCl_2$ -dppf, bis-(diphenylphosphinof-errocene)palladium dichloride, is then added, the mixture is heated to the boil, and 4.6 g of the product from step 1.1, dissolved in a little THF, are added dropwise. The reaction mixture is subsequently heated under reflux for 15 h. The batch is hydrolysed using water and subjected to extractive work-up. The crude product is purified by chromatography (pentane/silica gel). Colourless liquid.

Step 23b.3

2.4~g of 4-butylphenylacetylene are initially introduced in 30~ml of THF and cooled to -78° C. 14.3~ml of a 1 M solution of lithium bis(trimethylsilyl)amide in hexane are added dropwise to this solution and allowed to react at -78° C. for a further 1 h. 14.3~ml of a 1 M solution of methoxy-9-BBN are then added dropwise, and the mixture is stirred at -78° C. for a further 2 h. In a second apparatus, 1.0~g of the product from

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the last step, dissolved in 40 ml of THF, is initially introduced with the catalyst comprising 0.2 g of tris(dibenzylideneacetone)dipalladium and 0.35 g of 2-dicyclohexylphosphino-2',6'-dimethoxybiphenyl, and the reaction solution from the first reaction is added at room temperature. The mixture is heated at the boil for 15 h. The batch is hydrolysed using water and subjected to extractive work-up. The crude product is purified by chromatography (pentane/silica gel). Recrystallisation from pentane gives the purified title product.

Phase sequence: C 45 N 1801.

Substance Example 24b

$$C_4H_9$$
 C_4H_9 C_4H_9

The compound is prepared analogously to Example 23b. Phase sequence: C 118 N 222 I; $\Delta n=0.435$; $\Delta \epsilon=2.6$.

Substance Example 25b

$$C_4H_9$$
 C_2H_5 C_4H_8

The title compound is prepared analogously to Example 23b.

Phase sequence: C 41 N 161 I.

Substance Example 26b

$$C_3H_7$$
 $C \equiv C$ $C \equiv C$ C_4H_9

Synthesis Example 26b

26b.1) Synthesis of 1-iodo-4-bromonaphthalene

$$Br$$
 I_2
 THF
 Br
 I_2
 I

 $100 \mathrm{~g}$ (350 mmol) of 1,4-dibromonaphthalene are initially introduced in 11 of THF, cooled to -70° C., and 235 ml of n-BuLi (1.6 M in hexane, 370 mmol) are added dropwise. After 1 h, $103 \mathrm{~g}$ (406 mmol) of I_2 in 250 ml of THF are added

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dropwise, the mixture is stirred at -70° C. for a further 2 h, warmed to 0° C. and quenched by the addition of 50 ml (644 mmol) of aqueous NaHSO3 solution (w=39%). The phases are separated, and the aqueous phase is extracted once with MTB. The combined organic phases are washed with saturated sodium chloride soln., dried over sodium sulfate, filtered and evaporated in a rotary evaporator. The residue is purified by column chromatography (SiO2, heptane), and the further purification is carried out by recrystallisation from isopropanol, giving 1-iodo-4-bromonaphthalene as a yellow solid.

26b.2) Synthesis of 1-bromo-4-(4-n-propylphenylethynyl)naphthalene

$$I$$
 Br
 $Cu(I)I, PdCl_2(PPh_3)_2$
 NEt_3
 Br
 20
 Br
 25

15.3~g~(43.6~mmol)~of~1-iodo-4-bromonaphthalene and <math display="inline">7.25~g~(5.3~mmol)~of~4-n-propylphenylacetylene are initially introduced in $200~ml~of~NEt_3,\,170~mg~(0.9~mmol)~of~copper(I)~iodide~and~600~mg~(0.9~mmol)~of~bis(triphenylphosphine)~35~palladium(II)~chloride~are~added,~and~the~mixture~is~refluxed~for~30~minutes. The batch is cooled, water and heptane are~added,~and~the~phases~are~separated. The~organic phase~is~washed~with~saturated~sodium~chloride~soln.,~dried~over~sodium~sulfate,~filtered~and~evaporated~in~a~rotary~evaporator. 40 The residue is purified by column chromatography (SiO $_2$, heptane), and the further purification is carried out by recrystallisation from isopropanol.

26b.3) Synthesis of 1-(4-n-butylphenylethynyl)-4-(4-n-propylphenylethynyl)naphthalene

$$H_7C_3$$
 Br
 $Cu(I)I,$
 $PdCl_2(PPh_3)_2$
 NEt_3
 C_3H_7
 C_4H_9

2.35~g (6.3 mmol) of 1-bromo-4-(4-n-propylphenylethynyl)naphthalene and 1.33~g (8.4 mmol) of 4-n-butylphenylacetylene are initially introduced in 40 ml of NEt₃, 60 mg (0.3 mmol) of copper(I) iodide and 200 mg (0.3 mmol) of

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bis(triphenylphosphine)palladium(II) chloride are added, and the mixture is refluxed for 18 h.

The batch is cooled, water and heptane are added, and the phases are separated. The organic phase is washed with saturated ammonium chloride solution and subsequently with saturated sodium chloride soln., dried over sodium sulfate, filtered and evaporated in a rotary evaporator. The residue of compound (I) is purified by column chromatography (SiO₂, heptane), and the further purification is carried out by recrystallisation from isopropanol.

MS (EI): m/e (%)=426 (100, M⁺), 397 (11, [M-ethyl]⁺), 383 (16, [M-propyl]⁺), 354 (18, [M-ethylpropyl]⁺), 177 (14, [M-ethylpropyl]²⁺).

Phase sequence: C 78 N 191 I; Δ n=0.450; Δ ϵ =2.9.

Substance Example 27b

$$C_3H_7$$
— $C\equiv C$ — C_3H_7

The compound is prepared analogously to Example 26b. Phase sequence: C 108° C. N 194° C. I.

Substance Example 28b

The compound is prepared analogously to Example 26b. Phase sequence: C 63 $^{\circ}$ C. N 171.7 $^{\circ}$ C. I; Δn =0.435; $\Delta \epsilon$ =2.3.

Substance Example 29b

$$C_4H_9$$
 $C \equiv C$ $C \equiv C$

The compound is prepared analogously to Example 26b. Phase sequence: C 76° C. N 176.2° C. I; Δ n=0.427; Δ ϵ =2.4.

Substance Example 30b

$$C_5H_{11}$$
 $C \equiv C$ $C \equiv C$

The compound is prepared analogously to Example 26b. Phase sequence: C 100° C. N 162° C. I.

$$C_6H_{13} - C \equiv C - C \equiv C - C_6H_{1}$$

The compound is prepared analogously to Example 26b. Phase sequence: C 61° C. N 139° C. I.

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erties, in particular in the microwave region. The compound has a nematic phase and a clearing point (T(N,I)) of 114.5° C. Further physical properties at 20° C. are: $n_e(589.3~\text{nm}){=}1.8563;~\Delta n(589.3~\text{nm}){=}0.3250;~\varepsilon_{\parallel}(1~\text{kHz}){=}3.40;~\Delta\varepsilon(1~\text{kHz}){=}0.8~\text{and}~\gamma_1{=}2100~\text{mPa·s}$. The compound is suitable for applications in the microwave region and/or millimeter wave region, in particular for phase shifters.

TABLE 1a

	Proper	ties of the	compoun	d PTP(2)TP-6-3	3 at 30 GHz	
T/° C.	$\epsilon_r, $	ϵ_r, \perp	τ	$\tan\delta_{\epsilon,r,} $	$\tan\delta_{\epsilon,r,}\bot$	η
20	3.22	2.44	0.242	0.0029	0.0064	37.9

Substance Example 32b

$$C_4H_9$$
 $C \equiv C$ C_4H_9

This compound is prepared analogously to the preceding compounds in accordance with the right-hand reaction path of Scheme 15.

Phase sequence: C 85° C. N 202.4° C. I; Δ n=0.446; Δ ϵ =2.0.

TABLE 1b

		Propert	ies of the	compound	d PTP(2)TP-6-3	3 at 19 GHz	
	T/° C.	$\epsilon_r, \mid \mid$	ϵ_r, \perp	τ	$\tan\delta_{\epsilon,r,} $	$\tan\delta_{\epsilon,r,}\bot$	η
30	20	3.35	2.42	0.278	0.0029	0.0061	45.2

Substance Example 33b

$$C_6H_{13}$$
 $C \equiv C$ C_3H

The compound is prepared analogously to Example 32b.

Phase sequence: C 112° C. N 253.5° C. I; Δn =0.484; $\Delta \epsilon$ =4.1.

Substance Example 34b

In addition, the properties of the compound n-1-pentyl-4'-cyanobiphenyl (also called PP-5-N or CB 15) and the liquid-crystal mixture ZLI-4792 (product from Merck KGaA, Darmstadt, Germany) are investigated at 19 GHz. The results are summarised in Table 2.

$$C_4H_9$$
 $C \equiv C$
 $C = C$
 C_4H_9

60

This compound is prepared analogously to the preceding 55 compounds in accordance with the left-hand reaction path of Scheme 15.

Phase sequence: C 93° C. N 212.4° C. I; Δn =0.527; $\Delta \epsilon$ =0.8.

USE EXAMPLES

Comparative Examples 1 to 3

A liquid-crystalline substance having the abbreviation PTP (2)TP-6-3 is prepared by the method of Hsu, C. S., Shyu, K. 65 F., Chuang, Y. Y. and Wu, S.-T., Liq. Cryst., 27 (2), (2000), pp. 283-287, and investigated with respect to its physical prop-

TABLE 2

Comparison of the properties of the
various examples at 19 GHz and 20° C.

					$\tan \delta_{\epsilon}$	
Example	LC	€,-,	ϵ_r, \perp	τ	r, Max.	η
Comp. 1	P2-6-3*	3.35	2.42	0.278	0.0061	45.2
Comp. 2	PP-5-N ^{\$}	3.06	2.66	0.131	0.0273	4.8
Comp. 3	ZLI^{\S}	2.57	2.29	0.107	0.0126	8.5
1	M-1	2.65	2.30	0.130	0.0126	10.3
2	M-2	3.35	2.41	0.280	0.0066	42.2
3	M-3	2.65	2.30	0.133	0.0125	10.6

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TABLE 2-continued

		omparison ous examp				
Example	LC	$\epsilon_r, $	ϵ_r, \perp	τ	$ an \delta_{\epsilon}$ r, Max	η
4	M-4	3.35	2.41	0.281	0.0064	43.6
5	M-5	3.34	2.41	0.280	0.0078	35.9
6	M-6	t.b.d.	t.b.d.	t.b.d.	t.b.d.	t.b.d.
7	M-7	t.b.d.	t.b.d.	t.b.d.	t.b.d.	t.b.d.
8	M-8	t.b.d.	t.b.d.	t.b.d.	t.b.d.	t.b.d.
9	M-9	t.b.d.	t.b.d.	t.b.d.	t.b.d.	t.b.d.

Notes:

*P2-6-3 PTP(2)TP-6-3,

LC liquid crystal,

\$PP-5-N CB15,

§ZLI ZLI-4792 and

t.b.d. to be determined.

Example 1

A liquid-crystal mixture M-1 having the composition and properties as indicated in the following table is prepared.

No.	Abbreviation	c/%
1	PTP(2,5F)TP-6-3	10.0
2	CP-5-F	9.0
3	CP-6-F	7.2
4	CP-7-F	5.4
5	CCP-2-OT	7.2
6	CCP-3-OT	10.8
7	CCP-4-OT	6.3
8	CCP-5-OT	9.9
9	CCEP-3-OT	4.5
10	CCEP-5-OT	4.5
11	CPG-3-F	10.8
12	CPG-5-F	9.0
13	CGPC-3-3	1.8
14	CGPC-5-3	1.8
15	CGPC-5-5	1.8
Σ		100.0

 Physical properties
$\begin{split} T(N,I) &= 95^{\circ} \text{ C.} \\ \Delta n &(20^{\circ} \text{ C.}, 589.3 \text{ nm}) = 0.126 \\ \Delta \epsilon &(20^{\circ} \text{ C.}, 1 \text{ kHz}) = 4.9 \\ k_{11} &(20^{\circ} \text{ C.}) = \text{t.b.d. pN} \\ k_{33} k_{11} &(20^{\circ} \text{ C.}) = \text{t.b.d.} \\ V_0 &(20^{\circ} \text{ C.}) = \text{t.b.d. V} \\ \gamma_1 &(20^{\circ} \text{ C.}) = 169 \text{ mPa} \cdot \text{s} \end{split}$

t.b.d.: to be determined.

This mixture is very highly suitable for applications in the microwave region and/or millimeter wave region, in particular for phase shifters.

TABLE 3

		Properti	es of mixt	ure M-1 at 19 C	ЭНz	
T/° C.	$\epsilon_r, $	ϵ_r, \perp	τ	$\tan\delta_{\epsilon,r,} \mid \mid$	$\tan \delta_{\epsilon,r,}\bot$	η
20	2.65	2.30	0.130	0.0049	0.0126	10.3

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Example 2

A liquid-crystal mixture M-2 having the composition and properties as indicated in the following table is prepared.

		Composition Compound	
	No.	Abbreviation	c/%
10	1 2	PTP(2,5F)TP-6-3 PTP(2)TP-6-3	10.0 90.0
	Σ		100.0
15		Physical properties	
		$T(N, I) = 116^{\circ} C.$	

This mixture is very highly suitable for applications in the microwave region and/or millimeter wave region, in particular for phase shifters.

TABLE 4

			Propertie	es of mixt	re M-2 at 19 G	Hz	
5	T/° C.	ϵ_r ,	ϵ_r, \perp	τ	tan $\delta_{\epsilon, r, l}$	$\tan \delta_{\epsilon,r,}\bot$	η
	20	3.35	2.41	0.280	0.0028	0.0066	42.2

Example 3

A liquid-crystal mixture M-3 having the composition and properties as indicated in the following table is prepared.

No.	Abbreviation	c/%
1	PTP(1,6F)TP-6-3	10.0
2	CP-5-F	9.0
3	CP-6-F	7.2
4	CP-7-F	5.4
5	CCP-2-OT	7.2
6	CCP-3-OT	10.8
7	CCP-4-OT	6.3
8	CCP-5-OT	9.9
9	CCEP-3-OT	4.5
10	CCEP-5-OT	4.5
11	CPG-3-F	10.8
12	CPG-5-F	9.0
13	CGPC-3-3	1.8
14	CGPC-5-3	1.8
15	CGPC-5-5	1.8

55	Physical properties
	$T(N, I) = 102^{\circ} C.$
	$\Delta n (20^{\circ} \text{ C.}, 589.3 \text{ nm}) = 0.129$
	$\Delta \epsilon$ (20° C., 1 kHz) = 4.9
	$k_{11} (20^{\circ} C.) = t.b.d. pN$
	$k_{33}/k_{11} (20^{\circ} C.) = t.b.d.$
60	$V_0 (20^{\circ} C.) = t.b.d. V$

 $\gamma_1 (20^{\circ} \text{ C.}) = 168 \text{ mPa} \cdot \text{s}$

Notes:

t.b.d.: to be determined.

This mixture is very highly suitable for applications in the microwave region and/or millimeter wave region, in particular for phase shifters.

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142 TABLE 5 TABLE 7

			Properti	es of mixt	ure M-3 at 19 C	Нz					Properti	es of mixt	ıre M-5 at 19 C	Hz	
20 2.65 2.30 0.133 0.0045 0.0125 10.6 20 3.34 2.41 0.280 0.0029 0.0078 35.	T/° C.	$\epsilon_r, $	ϵ_r, ot	τ	$\tan\delta_{\epsilon,r,} $	$\tan\delta_{\epsilon,\;r,\;\perp}$	η	5	T/° C.	$\epsilon_r, \mid \mid$	ϵ_r , \perp	τ	$\tan \delta_{\epsilon,r,} $	$\tan \delta_{\epsilon,r,}\bot$	η
	20	2.65	2.30	0.133	0.0045	0.0125	10.6		20	3.34	2.41	0.280	0.0029	0.0078	35.9

Example 4

A liquid-crystal mixture M-4 having the composition and 15 properties as indicated in the following table is prepared.

	Composition Compound		20
No.	Abbreviation	c/%	
1 2	PTP(1,6F)TP-6-3 PTP(2)TP-6-3	10.0 90.0	
Σ		100.0	2:
	Physical properties		_
	$T(N, I) = 122^{\circ} C.$		
			3

This mixture is very highly suitable for applications in the microwave region and/or millimeter wave region, in particular for phase shifters.

TABLE 6

		Properti	es of mixt	ure M-4 at 19 C	Hz	
T/° C.	$\epsilon_r, \mid \mid$	ϵ_r, \perp	τ	$\tan\delta_{\epsilon,r,}\mid\mid$	$\tan\delta_{\epsilon,r,}\bot$	η
20	3.35	2.41	0.281	0.0025	0.0064	43.6

Example 5

A liquid-crystal mixture M-5 having the composition and properties as indicated in the following table is prepared.

No.	Compound Abbreviation	c/%	
1	PTP(1,6F)TP-6-3	15.0	_
2	PTP(2,5F)TP-6-3	20.0	
3	PTP(2)TP-6-3	65.0	
Σ		100.0	
	Physical properties		

This mixture is very highly suitable, with good low-temperature stability, for applications in the microwave region 65 and/or millimeter wave region, in particular for phase shifters.

Example 6

A liquid-crystal mixture M-6 having the composition and properties as indicated in the following table is prepared.

No.	Abbreviation	c/%
1 2	PTP(3 × 1)TP-4-4 PTP(2)TP-6-3	10.0 90.0
Σ		100.0
	Physical properties	
	$T(N, I) = 126^{\circ} C.$	

This mixture is very highly suitable for applications in the microwave region and/or millimeter wave region, in particular for phase shifters.

TABLE 8

			Propertie	es of mixt	ure M-6 at 19 G	Hz	
40	T/° C.	$\epsilon_r, $	ϵ_r, \perp	τ	tan $\delta_{\epsilon, r, +}$	$\tan\delta_{\epsilon,r,}\bot$	η
40	20	t.b.d.	t.b.d.	t.b.d.	t.b.d.	t.b.d.	t.b.d.

Note:

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t.b.d. to be determined.

Example 7

A liquid-crystal mixture M-7 having the composition and $_{50}$ properties as indicated in the following table is prepared.

		Composition Compound	
;	No.	Abbreviation	c/%
	1 2	PTP(4 × 1)TP-4-4 PTP(2)TP-6-3	10.0 90.0
	Σ		100.0
		Physical properties	
		$T(N, I) = 130^{\circ} C.$	

This mixture is very highly suitable for applications in the microwave region and/or millimeter wave region, in particular for phase shifters.

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t.b.d.

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Properties of mixture M-7 at 19 GHz								
T/° C.	$\epsilon_r, $	ϵ_r, \perp	τ	$\tan \delta_{\epsilon,r,} $	$\tan \delta_{\epsilon,r,}\bot$	η		

t.b.d.

t.b.d.

t.b.d.

20 Note:

t.b.d. to be determined.

t.b.d.

t.b.d.

Example 8

A liquid-crystal mixture M-8 having the composition and properties as indicated in the following table is prepared.

No.	Abbreviation	c/%
1	PTP(3 × 1)TP-4-4	10.0
2	CP-5-F	9.0
3	CP-6-F	7.2
4	CP-7-F	5.4
5	CCP-2-OT	7.2
6	CCP-3-OT	10.8
7	CCP-4-OT	6.3
8	CCP-5-OT	9.9
9	CCEP-3-OT	4.5
10	CCEP-5-OT	4.5
11	CPG-3-F	10.8
12	CPG-5-F	9.0
13	CGPC-3-3	1.8
14	CGPC-5-3	1.8
15	CGPC-5-5	1.8
Σ		100.0

Notes:

t.b.d.: to be determined.

This mixture is very highly suitable for applications in the microwave region and/or millimeter wave region, in particular for phase shifters.

 $T(N, I) = 102^{\circ} C.$

 Δ n (20° C., 589.3 nm) = 0.128 Δ e (20° C., 1 kHz) = 4.9

 $k_{11} (20^{\circ} \text{ C.}) = \text{t.b.d. pN}$

 $\begin{aligned} k_{33}/k_{11} & (20^{\circ} \text{ C.}) = \text{t.b.d.} \\ V_{0} & (20^{\circ} \text{ C.}) = \text{t.b.d. V} \\ \gamma_{1} & (20^{\circ} \text{ C.}) = 170 \text{ mPa} \cdot \text{s} \end{aligned}$

TABLE 10

Properties of mixture M-8 at 19 GHz						
T/° C.	$\epsilon_r, $	ϵ_r, \perp	τ	$\tan\delta_{\epsilon,r,}\mid\mid$	$\tan\delta_{\epsilon,r,}\bot$	η
20	t.b.d.	t.b.d.	t.b.d.	t.b.d.	t.b.d.	t.b.d.

Note:

t.b.d. to be determined.

Example 9

A liquid-crystal mixture M-9 having the composition and properties as indicated in the following table is prepared.

Composition Compound					
No.	Abbreviation	c/%			
1	PTP(4 × 1)TP-4-4	5.0			
2	CP-5-F	9.5			
3	CP-6-F	7.6			
4	CP-7-F	5.7			
5	CCP-2-OT	7.6			
6	CCP-3-OT	11.4			
7	CCP-4-OT	6.7			
8	CCP-5-OT	10.5			
9	CCEP-3-OT	4.7			
10	CCEP-5-OT	4.7			
11	CPG-3-F	11.4			
12	CPG-5-F	9.5			
13	CGPC-3-3	1.9			
14	CGPC-5-3	1.9			
15	CGPC-5-5	1.9			
Σ		100.0			

Physical properties $T(N, I) = 99^{\circ} C.$ $\Delta n (20^{\circ} C., 589.3 \text{ nm}) = 0.113$ $\Delta \epsilon (20^{\circ} C., 1 \text{ kHz}) = 5.2$ $k_{11} (20^{\circ} C.) = \text{t.b.d. pN}$ $k_{33}/k_{11} (20^{\circ} C.) = \text{t.b.d.}$ $V_{0} (20^{\circ} C.) = \text{t.b.d. V}$ $\gamma_{1} (20^{\circ} C.) = 148 \text{ mPa} \cdot \text{s}$

Notes

t.b.d.: to be determined.

This mixture is very highly suitable for applications in the microwave region and/or millimeter wave region, in particular for phase shifters.

TABLE 11

	Properties of mixture M-9 at 19 GHz						
	T/° C.	$\epsilon_r, \mid \mid$	ϵ_r , \perp	τ	$\tan\delta_{\epsilon,r,} \mid \mid$	$\tan\delta_{\epsilon,r,}\bot$	η
)	20	t.b.d.	t.b.d.	t.b.d.	t.b.d.	t.b.d.	t.b.d.

Note:

t.b.d. to be determined.

The invention claimed is:

1. Liquid-crystal medium which comprises a component A which comprises one or more compounds selected from the compounds of the formulae IA-1 to IA-3

IA-1

$$R^{11}$$
 $C \equiv C$ R^{13} $C \equiv C$ R^{12} $IA-2$

$$R^{11}$$
 $C \equiv C$ R^{13} $C \equiv C$ R^{12}

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IA-3

-continued

$$R^{11}$$
 $C \equiv C$ R^{13} $C \equiv C$ R^{12}

in which

 L^1 denotes halogen, R^{11} to R^{13} , independently of one another, denote unfluorinated alkyl or unfluorinated alkoxy, each having 1 to 15 C atoms, unfluorinated alkenyl, unfluorinated alkenyloxy or unfluorinated alkoxyalkyl, each having 2 to 15 C atoms, or cycloalkyl, alkylcycloalkyl, cycloalkenyl, alkylcycloalkenyl, alkylcycloalkylalkyl or alkylcycloal kenylalkyl, each having up to $15\,\mathrm{C}$ atoms, provided that R¹³ is not unfluorinated alkoxy, and which optionally comprises one or more compounds of the formula

$$R^{11}$$
 $C \equiv C$ A^{1} $C \equiv C$ R^{12} R^{12}

in which

$$A^{1}$$
 35

denotes one of the following groups:

25

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-continued CH₃

CH₃

$$CH_3$$
 $R^{13'}$
 $R^{14'}$
 $R^{14'}$
 $R^{14'}$
 $R^{14'}$
 $R^{14'}$
 $R^{14'}$
 $R^{14'}$

 $R^{11^{\circ}}$ to $R^{14^{\circ}}$, independently of one another, denote unfluorinated alkyl or unfluorinated alkoxy, each having 1 to 15 C atoms, unfluorinated alkenyl, unfluorinated alkenyloxy or unfluorinated alkoxyalkyl, each having 2 to 15 C atoms, or cycloalkyl, alkylcycloalkyl, cycloalkenyl, alkylcycloalkenyl, alkylcycloalkylalkyl or alkylcycloalkenylalkyl, each having up to 15 C atoms, and alternatively one of $R^{13^{\circ}}$ and $R^{14^{\circ}}$ or both also denote H. $_{40}$

2. Liquid-crystal medium according to claim 1, wherein component A comprises one or more compounds of the formula IB which are selected from the group of the compounds of the formulae IB-1 and IB-3 to IB-4

wherein L¹ and X¹ are selected to provide a ring A¹ as defined in claim 1 and in which the other parameters have the meanings given in claim 1 under formula IB.

 Liquid-crystal medium according to claim 2, wherein component A comprises one or more compounds of the formula IB-3 and/or of the formula IB-4, as indicated in claim 2.

4. Liquid-crystal medium according to claim 2, wherein component A comprises one or more compounds of the formula IB-1, as indicated in claim 2, in which X¹ denotes H.

5. Liquid-crystal medium according to claim **2**, wherein component A comprises one or more compounds of the formula IB-1, as indicated in claim **2**, in which X¹ denotes F.

6. Liquid-crystal medium according to claim 1, wherein, besides a component A, it additionally comprises one or more components selected from the group of the following components B to E:

a strongly dielectrically positive component, component B, which has a dielectric anisotropy of 10 or more,

a strongly dielectrically negative component, component C, which has a dielectric anisotropy having a value of -5.0 or less.

a component, component D, which has a dielectric anisotropy in the range from more than -5.0 to less than 10.0 and consists of compounds having seven or more five- or six-membered rings, and

a component, component E, which likewise has a dielectric anisotropy in the range from more than -5.0 to less than 10.0 and consists of compounds having one to six five-or six-membered rings.

7. Liquid-crystal medium according to claim 6, which comprises a component B.

8. Liquid-crystal medium according to claim **6**, which comprises a component C.

9. Liquid-crystal medium according to claim **6**, which comprises a component D.

10. Liquid-crystal medium according to claim 1, which further comprises one or more compounds of the formula VI

$$R^{11'} \longrightarrow C \equiv C \longrightarrow R^{12'}$$

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$$L^{61} \underbrace{\hspace{1.5cm}}^{\hspace{1.5cm}} A^{61} \underbrace{\hspace{1.5cm}}^{\hspace{1.5cm}} Z^{61} \underbrace{\hspace{1.5cm}}^{\hspace{1.5cm}} A^{62} \underbrace{\hspace{1.5cm}}^{\hspace{1.5cm}} Z^{62} \underbrace{\hspace{1.5cm}}^{\hspace{1.5cm}} A^{63} \underbrace{\hspace{1.5cm}}^{\hspace{1.5cm}} L^{62}$$

in which

 L^{61} denotes R^{61} and, in the case where Z^{61} and/or Z^{62} $_{10}$ denote trans-CH—CH— or trans-CF—CF—, alternatively also denotes X^{61} ,

L⁶² denotes R⁶² and, in the case where Z⁶¹ and/or Z⁶² denote trans-CH—CH— or trans-CF—CF—, alternatively also denotes X⁶²,

R⁶¹ and R⁶², independently of one another, denote H, unfluorinated alkyl or unfluorinated alkoxy having 1 to 17 C atoms or unfluorinated alkenyl, unfluorinated alkenyloxy or unfluorinated alkoxyalkyl having 2 to 15 C atoms,

x⁶¹ and X⁶², independently of one another, denote F or Cl, —CN, —NCS, —SF₅, fluorinated alkyl or alkoxy having 1 to 7 C atoms or fluorinated alkenyl, alkenyloxy or alkoxyalkyl having 2 to 7 C atoms, or —NCS,

one of Z⁶¹ and Z⁶² denotes trans-CH—CH—, trans-CF—CF— or —C—C— and the other, independently thereof, denotes trans-CH—CH—, trans-CF—CF— or a single bond, and

$$A^{61}$$
 to A^{63} ,

independently of one another, denote

150

-continued

11. Liquid-crystal medium according to claim 1, which comprises a compound of the formula IB.

12. Process for the preparation of a liquid-crystal medium according to claim 1, comprising mixing one or more compounds of the formula IA, as indicated in claim 1, with one or more further compounds and/or with one or more additives.

13. Component for high-frequency technology, which contains a liquid-crystal medium according to claim 1.

14. Microwave antenna array, which comprises one or more components according to claim 13.

15. Process for tuning a microwave antenna array, comprising electrically addressing a component according to claim 13.

16. Compound of one of the formulae IA-1 to IA-3:

 $R^{11} - C = C - C - R^{13}$ $C = C - R^{12}$

 $C \equiv C$ $C \equiv C$

IA-2

$$R^{11} - C \equiv C - CH_3 - C \equiv C - R^{12}$$

in which

L¹ denotes halogen,

R¹¹ to R¹³, independently of one another, denote unfluorinated alkyl or unfluorinated alkoxy, each having 1 to 15 C atoms, unfluorinated alkenyl, unfluorinated alkenyloxy or unfluorinated alkoxyalkyl, each having 2 to 15 C atoms, or cycloalkyl, alkylcycloalkyl, cycloalkenyl, alkylcycloalkylalkyl or alkylcycloalkenylalkyl, each having up to 15 C atoms, provided that R¹³ is not unfluorinated alkoxy.

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